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6210. PRIMARY CIRCULAR SUTURE OF THE ARTERIES IN INFECTED WOUNDS (Russian text) - Sichenikov I. A. - KHIRURGIYA 1958, 6 (98-106) Tables 1 Illus. 3

Experiments were carried out on 74 dogs; 62 experiments with administration of antibiotics and 12 without antibiotics. Arteries were sutured mechanically in 35 cases and manually in 39. Primary circular suture in suppurating wounds gave favourable results in the first group. Best results were obtained if surgical treatment of infected wounds was carried out within 24 hr. after their infliction. The results depend to a great extent on the rational choice of antibiotics, based on analysis of the wound microflora. Mechanical suture has considerable advantages in infected wounds over the manual suture. The results of these experiments make it possible to recommend wider application of primary suture of arteries, even in conditions with pronounced inflammation.

## NAME &amp; BOOK INFORMATION

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and M.O. Shchegoleva, Candidate of Economics.

**PURPOSE:** This collection of papers is intended to furnish information on industrial resources in Eastern Siberia and to provide a basis for future developmental planning in the field of ferrous metallurgy.

**CONTENTS:** The collection is a summary of the proceedings of the Ferrous Metallurgy Section of the First Conference of Representatives of the Academy of Sciences USSR, the State Planning Commission and the Council of Ministers USSR on the Development of the Industrial Resources of Eastern Siberia. The collection deals with four main areas of development in Eastern Siberia: 1) Mineral Resources; 2) the Coal base; 3) prospects for the development of ferrous metallurgy; and 4) problems in the development of aluminum-magnesium. A list of the 112 authors of the Section with their affiliations is given in the Appendix. References accompany several of their articles.

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SICHENKO, V.K.; IVANOV, B.V.; POLYAKOV, I.I.; REZNIKOV, A.A.;  
DORFMAN, G.A.; IZRAELIT, E.M.; NOTYCH, A.G.; TOPYGIN,  
L.A.; CHALYY, G.Ya.; STETSENKO, Ye.Ya.; UDOVICHENKO, L.V.;  
FILIPPOV, B.S., nauchn. red.; LERNER, R.Z., nauchn. red.;  
GOL'DIN, Ya.A., glav. red.; KULESHOV, M.M., red.; POLOTSK,  
S.M., red.

[By-product coke industry] Koksokhimicheskoe proizvodstvo.  
Moskva, Metallurgiya, 1965. 167 p. (MIRA 18:7)

1. Tsentral'nyy nauchno-issledovatel'skiy institut in-  
formatsii i tekhniko-ekonomiceskikh issledovaniy chernoy  
metallurgii. 2. Direktor Tsentral'nogo nauchno-issledova-  
tel'skogo instituta informatsii i tekhniko-ekonomiceskikh  
issledovaniy chernoy metallurgii (for Kuleshov).

SCHIFFER, J.

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A

Synthesis of pseudoconhydrine. I. Synthesis of one isomeride of racemic 2-propyl-5-hydroxypiperidine. E. Sorm and J. Sicher. *Collection Czechoslov. Chem. Compounds*, 14, 331-341 (1949) (in English). Oxidation of 2-alkyl-8-hydroxyquinolines with fuming  $H_2SO_4$  in the presence of  $V_2O_5$  gives 2-alkylquinoline acids from which a substituted 2-alkylpyridines and piperidines can be prep'd. Alkylation of 8-methoxyquinoline with  $PtCl_6$  gives 2-propyl-8-methoxyquinoline (I),  $b_p$  116-21° (picrate, m. 152-3°). Refluxing with 10%  $HgBr$  formed 2-propyl-8-hydroxyquinoline (II),  $b_p$  .. 80-91° (picrate, m. 155-6°); benzoate, m. 73-4°. Oxidation gave 31%, 2-propyl-5,6-pyridinedicarboxylic acid (III), m. 142-3°, and 2-propyl-8-pyridone-5-carboxylic acid, m. 161°. Decarboxylation in boiling  $EtCO_2H$  gave 2-propyl-5-pyridinecarboxylic acid, m. 120-30°, which was converted via the Et ester,  $b_p$  135-6°, hydrazide, m. 90-1°, azide, and urethan, m. 70°, to 2-propyl-5-aminopyridine,  $b_p$  134-6° (picrate, m. 161°). Diazotization yielded 2-propyl-5-hydroxypyridine, m. 91-2°, which was hydrogenated ( $Pt$  catalyst) to *d*-2-propyl-5-hydroxypiperidine, m. 90-1°. Oxidation of 2-propylquinoline (IV) with 10%  $HgBr$  in  $H_2SO_4$  and  $CuSO_4$  formed 2,3,4-pyridinetricarboxylic acid and not the expected III. II can be obtained from IV via the sulfonic acid in poor yields. John Howe Scott

CP

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Comments on the condensation of  $\rho$ -nitrobenzoyl chloride with maleic and cyanoacetate. M. Svoboda, J. Sicher, J. Gut, and F. Hrabák (Tech. Univ., Prague, Czech.) *Chem. Listy* 46, 51-2 (1952).—An ether soln. (100 ml.) contg. 9.23 g.  $\rho$ -O<sub>2</sub>N-C<sub>6</sub>H<sub>4</sub>-COCl (I), 8 g. C<sub>4</sub>H<sub>6</sub>(CO<sub>2</sub>Et)<sub>2</sub>, and 1.15 g. Na in 10 ml. EtOH was allowed to stand 4 days at room temp., then acidified with dil. H<sub>2</sub>SO<sub>4</sub>, treated with Na<sub>2</sub>CO<sub>3</sub> soln., and evapd.; a compd., m. 94-5° (from MeOH), suggested to be ( $\rho$ -O<sub>2</sub>N-C<sub>6</sub>H<sub>4</sub>-CO)<sub>2</sub>C(CO<sub>2</sub>Et)<sub>2</sub> or perhaps ( $\rho$ -O<sub>2</sub>N-C<sub>6</sub>H<sub>4</sub>-CO)( $\rho$ -CO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>-CO)<sub>2</sub>C(CO<sub>2</sub>Et)<sub>2</sub>, was obtained in 2.06 g. yield. Acidifying the Na<sub>2</sub>CO<sub>3</sub> ext. pptd. 2.45 g.  $\rho$ -O<sub>2</sub>N-C<sub>6</sub>H<sub>4</sub>-COCl(CO<sub>2</sub>Et)<sub>2</sub>, m. 43-5°. Similar reaction occurred between I, 6 g., and Na cyanoacetate (from 3.3 g. of the ester) and gave 4 g. ( $\rho$ -O<sub>2</sub>N-C<sub>6</sub>H<sub>4</sub>-CO)<sub>2</sub>C(CN)CO<sub>2</sub>Me, m. 153-4° (from MeOH), and 1.7 g.  $\rho$ -O<sub>2</sub>N-C<sub>6</sub>H<sub>4</sub>-COCH(CN)CO<sub>2</sub>Me, m. 157° (from MeOH), also obtained in 35% yield from EtOMgCH(CN)CO<sub>2</sub>Me and I.

M. Hudlický

SICHER, J.

SICHER, J.; FARKAS, J.; SORM, F.

Studies in the chloramphenicol series. Part 4. Synthesis of 1-(p-nitro-phenyl)-2-hydroxymethyl-2-dichloro-acetamido-1,3-propanediol and a correction [in English with summary in Russian]. Sbor.Chekh.khim.rab. 18 no.1:102-105 F '53. (MIRA 7:6)

1. Chentral Chemical Research Institute, Prague.  
(Chloramphenicol)

Sicher, Jiri

✓Chloramphenicol series. V. Analogs containing chlorine in the side chain and oxazolines. Jiri Lukáš and Jiri Sicher (Czech. akad. ved., Prague). Collection Czechoslov. Chem. Commun., 18, 460-80 (1963) (in English).—See C.A. 49, 218b. VI. Steric course of the reduction of dehydrochloramphenicol and related compounds. Jiri Sicher, Miroslav Svoboda, Magdalena Hrdá, Josef Rudičer, and František Šorm (Czech. akad. ved., Prague). Ibid. 487-99.—See C.A. 49, 319b. H. L. H.

SICHER, J., FARKAS, J.

Sicher, J., Farkas, J. "Studies in the chloramphenical series." V. Analogues containing chlorine in the side chain and oxazolines. p. 552 CASOPIS PRO PESTOVANI MATEMATIKY. CZECHOSLOVAK MATHEMATICAL JOURNAL. Vol. 47, no. 4, Apr. 1953, Praha, Czechoslovakia.

SO: Monthly List of East European Accessions, LC., Vol. 3, No. 1, Jan. 1954, Uncl.

SICHER, J. and others

Sicher, J. and others "Studies in the chloramphenicol series." VI. The steric course of the reduction of dehydrochloramphenicol and of related compounds. p. 565 CASOPIS PRO PESTOVANI MATEMATIKY. CZECHOSLOVAK MATHEMATICAL JOURNAL. Vol. 47, no. 4, Apr. 1953, Praha, Czechoslovakia

SO: Monthly List of East European Accessions, LC., Vol. 3, No. 1, Jan. 1954, Uncl.

SICHER, J.

SICHER, J.; SVOBODA, M.; FARKAS, J.; SORM, F.

Studies in the chloramphenicol series. Part 7. The side reactions  
in the reduction of dehydrochloramphenicol [in English with summary in  
Russian]. Sbor.Chekh.khim.rab. 19 no.2:317-329 Ap '54. (MLRA 7:6)

1. Department of Organic Synthesis, Institute of Organic Chemistry,  
Czechoslovak Academy of Science, Prague. (Chloromycetin)

FISHER, J.; PLAUS, J.; DRAZINA, A.

"Studies on the Chloramphenicol Series. VIII. Some Unsaturated Derivatives.

In English." p. 545,

(COLLECTION OF CZECHOSLOVAK CHEMICAL COMMUNICATIONS. SPORNÍK ČEJKHOŠLOVATSKÝH KHEMIČSKÝH PRÁCE, Vol. 19, No. 3, June 1954, Praha, Czechoslovakia)

SO: Monthly List of East European Accessions, (ISSN), LC, Vol. 4  
No. 5, May 1955, Uncl.

*Sicher, Jiri*  
*Czech*

Studies in the chloramphenicol series. IX. Side-chain  
 hetero-analog and thiazolinine. II. Early and J. Sicher (Czechoslovak Acad. Prague), *Chem. Listy* 46, 666 (1952); *Collection Czechoslov. Chem. Commun.* 19, 705 (1954). *Chem. Abstr.* 44, 53014. The prep. of  
*erythro-p,p'-bis(2-chlorophenyl)thio-CH<sub>2</sub>SH* (III). It is de-  
 scribed and the results of pharmac. tests of the chloro and  
 mercapto analogs of chloramphenicol given. Adding  
*p-O<sub>2</sub>NCH<sub>2</sub>COCHCl<sub>2</sub>CH<sub>2</sub>SH* (II), m. 132°,  
*p-O<sub>2</sub>NCH<sub>2</sub>COCH(NHCOCHCl<sub>2</sub>)CH<sub>2</sub>SH* (IV), m. 132°,  
*V* (from EtOH). Similarly, 1.21 g. III in 2 ml. dioxane treated  
 with 0.8 g. BrSH at 50° gave 1.4 g. *p-O<sub>2</sub>NCH<sub>2</sub>COCH-*  
*(CH<sub>2</sub>SH)<sub>2</sub>* (V), m. 184° (from dioxane). In  
 the same manner was prep'd. *p-O<sub>2</sub>NCH<sub>2</sub>COCH(NHAc)-*  
*CH<sub>2</sub>SH*, m. 140-1° (from EtOH), in 72% yield. Distg.  
 the mixt. of 3.79 g. IV, 70 ml. iso-PrOH, and 4.08 g. Al  
 (iso-PrO)<sub>3</sub> 3 hrs. under CO<sub>2</sub>, decomp., the residue with H<sub>2</sub>O  
 5 min. at 90°, extg. with Et<sub>2</sub>O, distg. off the solvent, and  
 crystg. the residue from EtOH yielded 2.03 g. (64%)  
*erythro-p-O<sub>2</sub>NCH<sub>2</sub>COH(CH<sub>2</sub>N(C(CHCl<sub>2</sub>)SCH<sub>2</sub>)VI*, m.

*132°*

The same compd. was obtained in 10% yield by the  
 Meerwein-Ponndorf reduction of V. Refluxing VI (1.28 g.)  
 in C<sub>6</sub>H<sub>6</sub> (5 ml.) with 1.52 g. *p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>Cl and 2 ml.  
 Me<sub>2</sub>N 2 hrs., dilg. the mixt. with 20 ml. Et<sub>2</sub>O, filtering off  
 the salt, evapg. the solvents, and adding petr. ether to the  
 residue gave 0.55 cryst. *erythro-p-O<sub>2</sub>NCH<sub>2</sub>CH(OSO<sub>3</sub>C<sub>6</sub>H<sub>4</sub>-  
 Me)CH<sub>2</sub>N(C(CHCl<sub>2</sub>)SCH<sub>2</sub>)*, m. 138° (from EtOH).

Heating 35.7 g. V with 33 g. Al (iso-PrO)<sub>3</sub> 2 hrs. with 480  
 ml. C<sub>6</sub>H<sub>6</sub>, decomp., the residue after distg. off the solvent  
 with a soln. of 10 g. tartaric acid in 100 ml. H<sub>2</sub>O, heating the

mix., 10 min. at 50°, dilg. with 200 ml. H<sub>2</sub>O, extg. with AcOEt, and distg. off the AcOEt gave 8.07 g. (22%) *ethoxy-p*-O<sub>2</sub>NC<sub>4</sub>H<sub>4</sub>CH(OH)CH<sub>2</sub>NHCOC(=O)CH<sub>2</sub>SBz<sub>2</sub> (VII), m. 207° (from EtOH). Using fresh liquid Al(iso-Pr)<sub>3</sub>, increased the yield to 49%. VII (1.0 g.) refluxed 15 min. with 200 ml. 10% AcOH yielded 1.28 g. (1.06 g. after cryst. from H<sub>2</sub>O) II, m. 159° (from PhMe). It was recrystallized to 71 by heating with Al(iso-Pr)<sub>3</sub> in *n*-PrOH (34 mg. out of 30 mg.). Treating 0.34 g. II in 50 ml. *n*-PrOH contg. 1 mol. C<sub>2</sub>H<sub>5</sub>N with 0.15 g. Br<sub>2</sub>Cl dissolved in 20 ml. H<sub>2</sub>O 30 min. at 15° gave 0.17 g. VII, m. 207°. *threo-p*-O<sub>2</sub>NC<sub>4</sub>H<sub>4</sub>CH(OH)CH<sub>2</sub>N(C(CH<sub>3</sub>)<sub>3</sub>O)CH<sub>2</sub> (C-4, 43; 3014) (VIII) (1.52 g.)

in 5 ml. dioxane added to Et<sub>2</sub>O sand, with HBr at -15°, dried after 5 min. with 15 ml. AcOH, the soln. washed with three 20-ml. portions H<sub>2</sub>O, dried, and evapd. yielded 0.82 g. *threo-p*-O<sub>2</sub>NC<sub>4</sub>H<sub>4</sub>CH(OH)CH<sub>2</sub>NHCOC(=O)CH<sub>2</sub>Br (IX), m. 146° (from 50% EtOH). Treating a soln. of 0.75 g. IX in 10 ml. Me<sub>2</sub>CO with a soln. contg. 0.21 g. NaSK in 10 ml. Me<sub>2</sub>CO yielded 0.21 g. *threo-p*-O<sub>2</sub>NC<sub>4</sub>H<sub>4</sub>CH(OH)CH(NHCOCH<sub>2</sub>CH<sub>2</sub>SBz<sub>2</sub>) m. 164° (from EtOH). Acetylation of IX with Ac<sub>2</sub>O and H<sub>2</sub>SO<sub>4</sub> gave 87% *threo-p*-O<sub>2</sub>NC<sub>4</sub>H<sub>4</sub>CHOAcCH(NHCOC(=O)CH<sub>2</sub>CH<sub>2</sub>SBz<sub>2</sub>) (X), m. 178° (from PhMe). Refluxing 8 g. X in C<sub>4</sub>H<sub>10</sub> 310 ml. with 1.5 g. P<sub>2</sub>S<sub>3</sub> 3 hrs. gave, after chromatographic purification, 1.84 g. (27%) *threo-p*-O<sub>2</sub>NC<sub>4</sub>H<sub>4</sub>CH(OAc)CH(NHCOC(=O)CH<sub>2</sub>CH<sub>2</sub>SBz<sub>2</sub>) (XI), m. 102-3° (from EtOH). Refluxing XI (0.51 g.) in 5 ml. EtOH with 1.21 g. Al(iso-Pr)<sub>3</sub> 3 hrs., distg. off the EtOH, heating the residue with 5 ml. 10 min. at 94°, distg. off the H<sub>2</sub>O, and crystallizing with Et<sub>2</sub>O yielded 0.41 g. *threo-p*-O<sub>2</sub>NC<sub>4</sub>H<sub>4</sub>CH(OAc)CH<sub>2</sub>N(C(CH<sub>3</sub>)<sub>3</sub>O)CH<sub>2</sub> (C-4, 43; 3014) (VIII) (1.52 g.)

$\text{CH}(\text{OH})\text{CH}_2\text{N}(\text{C}(\text{CH}_3)_2)\text{S}.\text{CH}_3$  (XII), m. 131° (from  $\text{CH}_3\text{COCl}$ ). Heating the mixt. of 0.32 g. XII, 10 ml.  $\text{AcOH}$ , and 18 ml.  $\text{H}_2\text{O}$  30 min. on the steam bath under  $\text{N}_2$ , dig. the soln. with 20 ml.  $\text{H}_2\text{O}$ , filtering the turbid soln. with  $\text{CaCO}_3$ , and evap., the filtrate *in vacuo* gave 0.08 g. I, m. 108°. Heating 0.36 g. XI 5 min. at 100° in 15 ml. 50%  $\text{HCO}_2\text{H}$  and dig. the mixt. with  $\text{H}_2\text{O}$  gave 0.25 g. (10.16 g. after crystall.) *trans*- $\text{p}$ - $\text{CH}_2\text{CH}(\text{OAc})\text{CH}(\text{NHCOCH}_3)\text{CH}_2$  XII, m. 165° (from XII). Refluxing 8.07 g. VII with 35 ml.  $\text{SOCl}_2$  1.5 hrs., distg. off the  $\text{SOCl}_2$  *in vacuo*, and crystg. the residue from  $\text{Et}_2\text{O}$  gave 5.53 g. *erythro*- $p$ - $\text{O}_2\text{NCH}_2\text{CH}(\text{NHCOCH}_3)\text{CH}_2\text{SBz}$ , (XIII), m. 132°. Refluxing 5.35 g. XIII with 50 ml.  $\text{BuOH}$  4 hrs. and distg. off the  $\text{BuOH}$  yielded 1.25 g. *threo*- $p$ - $\text{O}_2\text{NCH}_2\text{CH}(\text{OAc})\text{CH}(\text{NHCOCH}_3)\text{CH}_2\text{SBz}$ , m. 164° (from EtOH). Refluxing 4.46 g. *erythro*- $p$ - $\text{O}_2\text{NCH}_2\text{CH}(\text{NHCOCH}_3)\text{CH}_2\text{OBz}$  (X) 30 min. in 29 ml. dry collidine (170°) and filtering off the collidine hydrochloride deposited after the addn. of  $\text{Et}_3\text{O}^+$  yielded 2.9 g. *trans*- $p$ - $\text{O}_2\text{NCH}_2\text{CH}(\text{OAc})$   $(\text{CH}_3\text{Cl})_2\text{NCH}_2\text{OBz}$ , m. 122-4° (from EtOH). Re-

CZECH

*J-Terpene*. LXIII. Total synthesis of chamazulene. A simple general synthesis of 1,4,7-substituted azulenes. Jiří Novák, František Šorm, and Jiří Šicher (Czech. Akad. věd, Prague). *Chem. Listy* 48, 1649-52 (1954). Collection Czechoslov. Chem. Commun. 19, 1204-73 (1954) (in English); cf. C.A. 49, 9064e.—A general synthesis of 1,4,7-trialkyl-azulenes has been worked out based on the hydrogenation of a suitably substituted *α*-dihydroxybenzosuberene to the satd. diol, oxidation of the diol to the corresponding dicarboxylic acid, and cyclization to the desired azulene skeleton. *Chamazulene* was synthesized in this way.  $2,3-(MeO)_2C_6H_3COCl$  and  $Pr_2Cd$  gave 72%  $2,3-(MeO)_2C_6H_3COPr$  (I), b.p. 104°; *semicarbazone*, m. 154.5° (from MeOH). Refluxing 120 g. I, 61 g.  $Me_2NH \cdot HCl$ , 33 g.  $(CH_3O)_2$ , and 2.5 ml. concd. HCl in 450 ml. EtOH 4 hrs. with efficient stirring, adding 10 g.  $(CH_3O)_2$ , refluxing 8 more hrs., distg. off most of the EtOH, stirring the cryst. residue several times with  $Et_2O$ , dissolving the crystals in  $H_2O$ , liberating the bases with 400 ml. 10% NaOH, extg. the mixt. with  $Et_2O$ , concd. the ext. *in vacuo* to 80 ml., and treating the concentrate with dry HCl gave 66 g.  $2,3-(MeO)_2C_6H_3CO-CH_2EtCH_2NMe_2 \cdot HCl$  (II), m. 141° (from EtOH), and 65 g. recovered I. Treating 21.5 g. II in 80 ml.  $H_2O$  with 40 ml. 10% NaOH, extg. the liberated base with  $Et_2O$ , adding with cooling 20 g. MeI, evapg. the excess MeI and  $Et_2O$  *in vacuo*, stirring the crystals with 100 ml. EtOH, treating the mixt. with  $NaCH(CO_2Et)_2$  (prep'd. from 2.2 g. Na and 15 g.  $CH_3CO_2Et$ ) in 100 ml. EtOH, refluxing the mixt. until no more  $Me_2N$  escaped (8 hrs.), distg. off the EtOH, dissolving the NaI in  $H_2O$ , refluxing the crude diester  $2,3-(MeO)_2C_6H_3COCH_2CH_2CH(CO_2Et)_2$  with 40 ml. 10% NaOH 2 hrs., acidifying the mixt., decarboxylating the free acid by heating 30 min. at 170°, and distg. the crude acid *in vacuo* gave 13.6 g.  $2,3-(MeO)_2C_6H_3COCH_2CH_2CH_2CO_2H$  (III), b.p. 100°. Hydrogenation of III in 200 ml. AcOH over 3 g. 5% Pt-C at 80-90° gave 30.6 g.  $2,3,4-Me_3O_2C_6H_3CH_2CH_2CH_2CO_2H$  (IV), b.p. 181°. Cyclizing IV by heating 1 hr. at 70° with 750 g. poly-

Sicher Jiri

Stereochemical studies. Introductory remarks. Jiri Sicher

Sicher (Czech. Akad. věd, Prague). *Chem. Listy* 49, 1515-1517 (1955)(in Engl.).—A program for the following series of papers on stereochemistry of noncyclic compounds is displayed, and the fundamental terms are explained. I. Steric course of the acid-catalyzed cyclization of *N*-thiobenzoyl derivatives of 1,2-amino alcohols. A new stereospecific reaction. Jiri Farkas and Jiri Sicher. *Chem. Listy*, 49, 1320-9; *Collection Czechoslov. Chem. Commun.*, 20, 1391-1401(1955)(in English).—A new stereospecific reaction was found in the acid-catalyzed cyclization of *N*-thiobenzoyl derivs. of epimeric 1,2-amino alcs. The *threo*-epimers give exclusively *trans*-2-oxazolines, the *erythro*-epimers yield only 20-40% *cis*-2-oxazolines whereas the main reaction products are *trans*-2-thiazolines. The starting  $\text{RCI}(\text{OH})\text{CH}(\text{NH}_2)\text{R}'$  (I) were prep'd. according to the literature. Physical consts. are given for:  $\text{R}=\text{R}'=\text{Me}$ , *threo*, m. 18-19°,  $n_D^{20}$  1.4450; *erythro*, m. 42-3°;  $\text{R}=\text{Ph}$ ,  $\text{R}'=\text{CO}_2\text{Et}$ , *threo*, m. 81-2°; *erythro*, m. 84°;  $\text{R}=\text{R}'=\text{Ph}$ , *threo*, m. 126-8°; *erythro*, m. 103°. Hydrogenating 7 g. *erythro*-compd.,  $\text{R}=\text{Me}$ ,  $\text{R}'=\text{Ph}$  in 99% EtOH (210 ml.) contg. 0.2% ethanolic HCl, over 0.7 g. PtO<sub>2</sub> at 15°, extg. the alkalinized mixt. with ether, and treating the base with excess HCl in EtOH gave 3.1 g. I.HCl ( $\text{R}=\text{Me}$ ,  $\text{R}'=\text{C}_6\text{H}_5$ ) m. 189-90°. The thiobenzoyl derivs. of the amino alcs.,  $\text{RCH}(\text{OH})\text{CHR}'\text{NHCSPh}$ , were prep'd. in 3 ways: method A: 0.01 mole of the amino alc. in 10 ml.  $\text{C}_6\text{H}_5\text{N}$  was heated 1 hr. at 100° with 0.011 mole PhCSSCH<sub>2</sub>-CO<sub>2</sub>H, the mixt. dild. with 100 ml. Et<sub>2</sub>O, the ether layer extd. with 5% HCl, with a satd. soln. of NaHCO<sub>3</sub>, the ether distd. off, and the residue stirred with petr. ether and recrystd. Method B: A soln. of 0.01 mole PhSSCH<sub>2</sub>CO<sub>2</sub>H in 15 ml. 2N NaOtf was added to 0.01 mole of the amino alc. or its HCl salt, the mixt. allowed to stand 5 hrs. at 20°, the thiobenzoyl deriv. filtered off and crystd. Method C: a 2-oxazoline deriv. (0.02 mole) was added to 25 ml. of a soln. of (NH)<sub>2</sub>S in MeOH, the mixt. allowed to stand 10

days, dild. with 500 ml. H<sub>2</sub>O, extd. with Et<sub>2</sub>O, the Et<sub>2</sub>O evapd., the residue distd. with C<sub>6</sub>H<sub>6</sub> and crystd. The compds. of the general formula  $\text{RCH}(\text{OH})\text{CHR}'\text{NHCSPh}$  are given (R, R' configuration, method of prep'n., yield in %, and m.p.): Me, Me, *threo* (I), *B*, 42, 80-1° (from C<sub>6</sub>H<sub>6</sub>-petr. ether); Me, Me, *erythro* (II), *B*, 54, 85° (from C<sub>6</sub>H<sub>6</sub>-petr. ether); Ph, Me, *threo* (III), *A*, *C*, 23% and 50%, resp., 79-80° (from C<sub>6</sub>H<sub>6</sub>-petr. ether); Ph, Me, *erythro* (IV), *A*, 67, 138° (from C<sub>6</sub>H<sub>6</sub>) [ $\text{PhC}(\text{SCH}_2\text{CO}_2\text{H})\text{CHMeNH}_2$ ] salt of *erythro*- $\text{PhCH}(\text{OH})\text{CHMeNH}_2$ , m. 141-2°; Me, Ph, *threo* (V), *C*, 32, 115-18° (from EtOH); Me, Ph, *erythro* (VI), *A*, 30, 132° (from EtOH); Ph, CO<sub>2</sub>Et, *threo* (VII), *A*, 46, 82-90° (from C<sub>6</sub>H<sub>6</sub>-petr. ether); Ph, CO<sub>2</sub>Et, *erythro* (VIII), *A*, 61, 141-4° (from EtOH); Ph, Ph, *threo* (IX), *C*, 65, 158° (from EtOH) [this reaction required 30 days for completion; otherwise, an addn. compd. of the oxazoline and the product (1:1), m. 118-20°, is isolated]; Ph, Ph, *erythro* (X), *A*, 46, 134-6° (from EtOH); cyclohexyl, Me, *erythro* (XI), *A*, 77, 141° (from 50% EtOH); Me, cyclohexyl, *erythro* (XII), *A*, 45, 92° (from C<sub>6</sub>H<sub>6</sub>-petr. ether). Cyclization of Ia-XII was carried out as follows: 0.003 mole Ia-XII was heated with 5 ml. 10% HCl in EtOH in a sealed tube at 100° until the yellow color disappeared. Two thirds of the EtOH was distd. off at 15 mm., and the residue was treated with Et<sub>2</sub>O. With III, V, VII, IX, and XI, cryst.  $\text{RCH}(\text{OBz})\text{CHR}'\text{NH}_2\text{Cl}$  (XIII) deposited, the mother liquors being the corresponding thiazolines were alkalinized with 1 eq. NH<sub>3</sub>, extd. with petr. ether, the petr. ether evapd., and the residual thiazoline distd. From the pq. layer after the extn. of the thiazolines with petr. ether small量. of  $\text{RCH}(\text{OH})\text{CHR}'\text{NH}_2\text{Bz}$  crystd. on standing. Transformation of the *threo* derivs. to XIII (compd., hrs. of reaction, yield in %, m.p. of XIII): III, 2, 62, 198-200° (from EtOH-H<sub>2</sub>O); back acyl

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migration gave *threo*- $\text{PhCH}(\text{OH})\text{CHMeNH}_2$ , m. 127°; V: 2, 57, 190-1° (from 5% HCl), back acyl migration gave *threo*- $\text{MeCH}(\text{OH})\text{CHPhNH}_2$ , m. 147° (from EtOH); VII: 2, 41, 190-1° (from 5% HCl); IX: 2, 73, 210° (from EtOH-H<sub>2</sub>O); I gave *trans*- $\text{MeCH}_2\text{CHMe.N:CPH}_2\text{O}$  (XIV)

HCl salt, 1 hr., 64%, m. 133-4° (from EtOH-Et<sub>2</sub>O). The cyclization of the *erythro* derivs. gave  $\text{RCH}_2\text{CHR}'\text{N:CPH}_2\text{S}$

(XIVa) (compd., time of heating, yield in %, b.p. of XIVa, m.p. of the picrates): II, 1, 62,  $b_{10}$  155-8°, 133° (from MeOH); IV, 5, 67,  $b_{10}$  162-7°, 141° (from EtOH); VI, 3, 50,  $b_{10}$  135-40°, 141° (from MeOH); in addn., 24% of *erythro*- $\text{MeCH}(\text{OBz})\text{CHPhNH}_2\text{Cl}$ , m. 189-90° (from 5% HCl), was isolated; VIII, 5, 55, m. 65-7° (from petr. ether),  $b_{10}$  175°, —; X, 2, 78, m. 93° (from petr. ether), —, —; XI, 1, 63,  $b_{10}$  125-30°, 147-8° (from MeOH); XII, 1, 55,  $b_{10}$  130°, m. 109° (from MeOH);  $\text{MeCH}(\text{OH})\text{CHMeNH}_2$  and BzCl gave *erythro*- $\text{MeCH}(\text{OH})\text{CHMeNH}_2\text{Bz}$  (XV), m. 122°. Treating 1 g. XV at 0° with 3 ml. SOCl<sub>2</sub>, distg. off the SOCl<sub>2</sub> after standing 1 hr. at 15°, and crystg. the residue from EtOH-Et<sub>2</sub>O gave 0.95 g. HCl salt of XIV, m. 133-4° (from EtOH-Et<sub>2</sub>O); free base XIV, m. 120°; picrate, m. 148° (from EtOH). Treating 7.85 g. XI in 20 ml. C<sub>6</sub>H<sub>6</sub> with a soln. of 2.2 g. NaOH in 20 ml. H<sub>2</sub>O and a soln. of 7.7 g. BzCl in 15 ml. C<sub>6</sub>H<sub>6</sub> at 15° gave 11.2 g. *erythro*- $\text{C}_6\text{H}_5\text{CH}(\text{OH})\text{CHMeNH}_2\text{Bz}$  (XVI), m. 126° (from C<sub>6</sub>H<sub>6</sub>). Treating at 5° 11.2 g. XVI with 30 ml. SOCl<sub>2</sub>, letting the soln. stand 3 hrs. at 15°, distg. off at 15 mm., the SOCl<sub>2</sub>, treating the residue with 50 ml. H<sub>2</sub>O, neutralizing the soln. with 20% NaOH, extg. the oxazoline deriv. with Et<sub>2</sub>O, and distg. the ext. yielded 8.9 g. *trans*- $\text{C}_6\text{H}_5\text{CH}_2\text{CHMe.N:CPH}_2\text{O}$ ,

$b_{10}$  135-7°; picrate, m. 171° (from EtOH). II. Reaction of N-acyl derivatives of 1,2-amino thiol in acidic solution. Jiri Sicher and Miroslav Svoboda, *Chem. Listy* 49, 1330-5; *Collection Czechoslov. Chem. Commun.* 20, 1103-8 (1955) (in English).—The epimeric *threo*- and *erythro*- $\text{RCH}(\text{SAc})\text{CHR}'\text{NCOR}'$  (I) behave differently in acidic medium. After the formation of the common intermediate, a 2-hydroxythiazolidine deriv., the *threo*-epimers give exclusively derivs. of *trans*-2-thiazoline (IIa) whereas the *erythro*-epimers give, in addn. to *cis*-2-thiazoline (IIb),  $\text{RCH}_2\text{(SCOR')CH}_2\text{NR'Cl}$  (III). The N-acyl derivs. of I give substantially larger amounts of III than the Bz derivs. The different behavior is explained as follows: the formation of IIa from *trans*-2-hydroxythiazoline is not sterically impeded since the introduction of the double bond does not influence the mutual distance of R and R' in the *threo*-epimers. On the other hand, in *cis*-2-hydroxythiazoline, formed from the *erythro*-epimers, the introduction of the double bond results in increasing the steric interaction of R and R' which are brought closer to each other. The steric factors will, therefore, interfere with the formation of IIb. I (R' = Ph) gain, after the formation of IIa and IIb, a conjugation of the formed double bond with the aromatic ring. I (R' = Me) do not have the possibility of having such conjugated system, and therefore, the acyl migration resulting in III will prevail over the formation of IIa and IIb. Heating a soln. of *threo*- $\text{PhCH}(\text{SAc})\text{CHMeNH}_2\text{Bz}$  (460 mg.) in EtOH satd. with HCl 8 hrs. at 100° in a sealed tube, evapg. the soln. to dryness, dilg. with H<sub>2</sub>O, extg. with Et<sub>2</sub>O, and chromatographing the residue after the evapn. of the Et<sub>2</sub>O gave 320 ml. (benzene fraction) of *trans*- $\text{PhCH}_2\text{CHMe.N:CPH}_2\text{O}$ .

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N: CPh.S; *picrate*, m. 141-2° (from EtOH). Heating a soln. of 3.13 g. *erythro*-PhCH(SAc)CHMeNHBz in EtOH (10 ml.) contg. HCl 8 hrs. at 100°, filtering off 180 mg. of *erythro*-PhCH(SBz)CHMeNH<sub>2</sub>HCl, m. 190-2° (from EtOH-Et<sub>2</sub>O), evapg. the mother liquor to dryness, dilg. the residue with H<sub>2</sub>O, extg. the soln. with Et<sub>2</sub>O, and distg. the ext. gave 1.35 g. *cis*-2,5-diphenyl-4-methyl-2-thiazoline, b.p. 145°; *picrate*, m. 123-4° (from EtOH). Heating 1.38 g. *threo*-PhCH(SAc)CHPhNH(Bz) 1 hr. at 100° with 50 ml. EtOH satd. with HCl, distg. off the EtOH, adding 20 ml. H<sub>2</sub>O, extg. the mixt. with Et<sub>2</sub>O, evapg. the solvent, and H<sub>2</sub>O, extg. the residue gave 1.53 g. crude and 1.44 g. cryst. *trans*-2,4,5-triphenyl-2-thiazoline (IIIa), m. 93-4°, giving isomorphous crystals, m. 68-9°. Heating 11.5 g. *erythro*-PhCH(SAc)CHPhNHBz with 200 ml. EtOH satd. with HCl 6 hrs. on the steam-bath, distg. off the EtOH, adding 50 ml. H<sub>2</sub>O and 100 ml. Et<sub>2</sub>O, filtering off the crystals (1.68 g.) of *erythro*-PhCH(SBz)CHPhNH<sub>2</sub>HCl, m. 226° (from EtOH-Et<sub>2</sub>O), extg. the mother liquor with Et<sub>2</sub>O, evapg. the Et<sub>2</sub>O, and chromatographing the residue gave 6.29 g. crude and 5 g. cryst. *cis* isomer of (IIIa), m. 84-5°. Heating 3.3 g. *threo*-PhCH(SAc)CH(CO<sub>2</sub>Et)NHBz with 100 ml. EtOH satd. with HCl 6 hrs. on the steam-bath, distg. off the EtOH, dilg. the residue with H<sub>2</sub>O, and extg. the soln. with Et<sub>2</sub>O gave 2.0 g. of an oil which yielded on boiling with petr. ether 150 g. insol. *trans*-2,5-diphenyl-4-carboxy-2-thiazoline, m. 137-8° (from C<sub>6</sub>H<sub>6</sub>), and 1.62 g. sol. *trans*-2,5-diphenyl-4-carbethoxy-2-thiazoline (IV), m. 135-6° (from petr. ether). Similar treatment of 1.65 g. *erythro*-PhCH(SAc)CH(CO<sub>2</sub>Et)NHBz gave 350 mg. *cis* isomer (V) of IV, m. 93-4° (from Et<sub>2</sub>O), and IV (from mother

liquors after treatment with petr. ether). Heating 19 hrs. gave IV exclusively. IV was obtained by refluxing 1 hr. 0.52 g. *threo*-PhCH(SH)CH(NH)CO<sub>2</sub>Et with 0.3 g. PhCOEt:NH in 5 ml. CHCl<sub>3</sub>, dilg. the mixt. with 20 ml. Et<sub>2</sub>O and extg. 3 times with a small vol. H<sub>2</sub>O. Heating 0.40 mg. *threo*-PhCH(SAc)CHPh(NHAc) in 10 ml. EtOH satd. with HCl 4 hrs. on the steam-bath gave 570 mg. *trans*-2-methyl-1,5-diphenyl-2-thiazoline (VI), b.p. 132°. Similar treatment of *erythro*-PhCH(SAc)CHPhNHAc gave 34.5% *cis* isomer of VI, m. 97-8° (from petr. ether). Infrared spectra of IV and V and of the corresponding oxazolines are given. III. Steric course of the reaction of N-acyl derivatives of 1,2-amino alcohols with thionyl chloride. Jiri Sicher and Magdalena Pluhková. *Chem. Listy* 49, 1330-50; *Collection Czechoslov. Chem. Commun.* 20, 1409-25 (1955) (in English).—1,2-N-acylamino alcohols react with SOCl<sub>2</sub> according to the configuration to give either 2-oxaz-SOCl<sub>2</sub> according to the configuration to give either 2-oxazoline (with inversion), or a  $\beta$ -chloro amide (without inversion). *erythro*-PhCH(OH)CHMeNHCO<sub>2</sub>H, NO<sub>2</sub>-p (I) gave, after a short action of SOCl<sub>2</sub>, 2-(nitrophenyl)-4-methyl-5-phenyl-2-oxazoline (II) and on prolonged action of SOCl<sub>2</sub> by a secondary reaction *erythro*-PhCH(OH)CHMeNHCO<sub>2</sub>H, NO<sub>2</sub>-p (III). *threo*-PhCH(OH)CHMeNHCO<sub>2</sub>H, NO<sub>2</sub>-p (IV) yielded by the reaction with SOCl<sub>2</sub> 30% *threo*-PhCH(OH)CHMeNHCO<sub>2</sub>H, NO<sub>2</sub>-p (V) and probably III, hydrolysis of which gave 30% *threo*-PhCH(OOCOC<sub>2</sub>H, NO<sub>2</sub>-p)CHMeNH<sub>2</sub>HCl (VI). III and V were not formed by secondary hydrolysis of the corresponding 2-oxazolines since these are stable under the conditions used for the reaction with SOCl<sub>2</sub>. The direct hydrolysis of the reaction mixt. after SOCl<sub>2</sub> upon IV yielded a mixt. of *erythro*- and *threo*-amino alcohols with the ratio of *threo* to *erythro* = 3:2. The two following couples, *erythro*- and *threo*-MeCH-

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(OH)CHPhNHCO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p* (VII), VIII, and *erythro*- and *threo*-C<sub>6</sub>H<sub>5</sub>CH(OH)CHMeNHCO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p* (IX, X) were transformed, with inversion, to *trans*- and *cis*-2-(*p*-nitrophenyl)-4-phenyl-5-methyl-2-oxazoline (XI, XII), and *trans*- and *cis*-2-(*p*-nitrophenyl)-4-methyl-5-cyclohexyl-2-oxazoline (XIII, XIV). The different result of the reaction of SOCl<sub>2</sub> with IV is ascribed to the polar effect of the Ph group since the corresponding cyclohexyl deriv. X behaved normally. Hydrolysis of the  $\beta$ -chloro amides afforded the corresponding derivs. RCH(O<sub>2</sub>CC<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p*)CHR'NH<sub>2</sub>HCl having the inverted configuration. The amino alcohols were prep'd. from the appropriate isonitroso ketones by hydrogenation over Raney Ni, or by the Meerwein-Ponndorf reduction of *p*-nitrobenzamido ketones during which formation of a 2-oxazoline deriv. was observed. The *threo*-epimers were obtained from the *erythro*-epimers by the reaction with SOCl<sub>2</sub> followed by hydrolysis. The cyclohexyl derivs. were prep'd. from the Ph deriv. by hydrogenation over PtO<sub>2</sub>. *Erythro*-PhCH(NH)CHMeOH was transformed to *trans*- $\beta$ -methylstyrene oxide (XIVa) (infrared spectra of *cis* and *trans* epoxides are given). The 2-oxazolines of *cis* and *trans* configuration were prep'd. for identification purposes from *p*-nitrobenzimino ethers and the amino alcohols. Hydrogenation of 407 g. BzCMe<sub>2</sub>NOH in 7.5 l. EtOH over 160 g. Raney Ni at room temp. and 120-50 atm. gave 150 g. *erythro*-Ph(OH)CHMeNH<sub>2</sub> (XV), m. 104-5°. Refluxing 11.2 g. BzCHMeNH<sub>2</sub>HCl with 12 g. *p*-O<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>4</sub>COCl (XVI) in 150 ml. PhMe 6 hrs., distg. off the PhMe, and crystg. the residue from EtOH gave 11 g. BzCHMeNHCO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p* (XVII), m. 123-4°. Treating 37 g. HCl salt of XV in 300 ml. H<sub>2</sub>O with 16 g. NaOH in 200 ml. H<sub>2</sub>O and 33 g. XVI in 500 ml. C<sub>6</sub>H<sub>6</sub> 50 min. gave 55 g. crude and 50 g. pure I, m. 189-90° (from EtOH). The same compnd. was prep'd. by refluxing 2.5 hrs. 9 g. XVII with 9.2 g. (iso-PrO)<sub>2</sub>Al and 200 ml. iso-PrOH, by decompg. the residue after the evapn. of the solvent with 50 ml. H<sub>2</sub>O, by evapg. the mixt. and

extg. the residue with AcOEt; in a yield of 8 g. Dissolving 4.5 g. I in 15 ml. SOCl<sub>2</sub>, evapg. the soln., dissolving the residue in 50% EtOH in the presence of 8 ml. 6N HCl in EtOH, refluxing the soln. 5-10 min., and filtering with C gave on cooling 4 g. HCl salt of IV, m. 215° IV, (by alkalinization in a 55% yield), m. 170° (from 75% EtOH). Refluxing 1.5 g. IV with 10% HCl 8-10 hrs., extg. the soln. with Et<sub>2</sub>O, evapg. the aq. soln. to dryness, and crystg. from EtOH and Et<sub>2</sub>O gave 0.7 g. *threo*-PhCH(OH)CHMeNH<sub>2</sub>-HCl, m. 169-70°; free base (in 66-7% yield from the HCl salt in Et<sub>2</sub>O with dry NH<sub>3</sub>), m. 71° (from C<sub>6</sub>H<sub>6</sub>). Hydrogenation of PhC(=NOH)Ac over Raney Ni gave 63% *erythro*-PhCH(NH)CHMeOH (XVIII), m. 85°. Heating 2.2 g. PhCHAcNH<sub>2</sub>HCl with 2.4 g. XVI in 40 ml. PhMe 4-6 hrs. gave 2.8 g. PhCHAcNHCO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p* (XIX), m. 156-7°. XVIII and XVI gave 80% *erythro*-PhCH(NHCO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p*)-CHMeOH (XX), m. 154-5°. The same compnd. was prep'd. by the Meerwein-Ponndorf reduction of XXI in a yield of 68%; 0.6% of XII, m. 125°, was isolated as a by-product. The reaction of XX with SOCl<sub>2</sub> and the subsequent hydrolysis gave 82% of the HCl salt of *threo*-PhCH(NHCO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p*)-CHMeOH (XXI), m. 207-8°; freebase XXI, m. 153-4°. Hydrolysis of XXI gave 55% *threo*-PhCH(NH<sub>2</sub>)CHMeOH, m. 90.5-1.5°. The *erythro*-configuration of XVIII was proved by prep'g. PhCH(CHMeOH)NH<sub>2</sub>I, m. 178° (yield 66%) and by transforming it to XXIa. Hydrogenation of 10.5 g. of XV-HCl in 300 ml. 90% EtOH over 1 g. PtO<sub>2</sub> in the presence of 0.3 ml. 99% EtOH satd. at 0° with HCl (room temp., normal pressure) gave 8 g. *erythro*-C<sub>6</sub>H<sub>5</sub>CH(OH)CHMeNH<sub>2</sub>, m. 82-3° (XXXII). XIII and XVI gave 53% *erythro*-C<sub>6</sub>H<sub>5</sub>CH(OH)CHMeNHCO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*p*, m. 163-4° (from 75% EtOH). Similarly was prep'd., by hydrogenating the *threo*-PhCH(OH)CHMeNH<sub>2</sub>HCl, *threo*-C<sub>6</sub>H<sub>5</sub>CH(OH)CHMeNH<sub>2</sub> (yield 93%), m. 83-4° (from petr. ether). Nitrobenzoylation gave 82% of a crude compnd. contg. di-*p*-nitrobenzoyl deriv. It was treated

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with 100 ml. 1% HCl in EtOH 3 days, the monobenzyl deriv. was thus transformed to the HCl salt, which yielded 66% *threo*- $C_6H_5CH(OH)CHMeNHCOCH_2NO_2-p$  (XIII), m. 120-1° (from 50% EtOH). The same compd. was prep'd. by heating XIII with concd. HCl by way of its HCl salt, m. 200-1° (from EtOH-Et<sub>2</sub>O). The reaction of the *p*-nitrobenzyl derivs. of the amino alcohols was carried out by dissolving the compd. in a min. amt. of SOCl<sub>2</sub> (1-2 ml. per 1 g. of the compd.) at 18-25°, by treating the mixt. after a certain time with ice and AcOEt, by evapg. the AcOEt ext., and by crystg. the residue, contg. 10-15% of the HCl salt of the *O*-acyl deriv. Thus were prep'd. the following (starting acyl deriv.), time of action of SOCl<sub>2</sub> (in min.), the product, m.p., yield, and the yield of the comparative compd. prep'd. by heating on the steam-bath equinol. amounts of the amino alcohol and Et *p*-nitrobenzimidate given: I, 1, II, 77-8°, 78%; VI, 1, XI, 125°, 77%; VIII, 30, XII, 121-2°, 53%; IX, 120, XIII, 113-14°, 60%; X, 75°; X, 240, XIV, 124-5°, 67%; XV, 72°. Dissolving 1.5 g. IV in 3 ml. SOCl<sub>2</sub>, decomposing the mixt. after 1 min. with ice, extg. with C<sub>6</sub>H<sub>6</sub>, gave 31% V, 128-9° (from C<sub>6</sub>H<sub>6</sub>). The mother liquor contained 30% VI, m. 215° (from H<sub>2</sub>O). In another expt., the residue after the evap. of C<sub>6</sub>H<sub>6</sub> from the ext. (2.85 g.) was refluxed 6 hrs. with 80 ml. and 2 ml. 3N HCl, the soln. evapd. to dryness, the residue extd. with hot C<sub>6</sub>H<sub>6</sub> gave 0.2 g. *erythro*- and *threo*-amides, and undissolved 2.08 g. of the *threo*- and *erythro*-PhCH(*O*<sub>2</sub>*CC*<sub>6</sub>*H*<sub>5</sub>*NO*<sub>2</sub>-*p*)CH<sub>2</sub>NH<sub>2</sub>HCl which were transformed by acyl migration to 1.13 g. I, m. 189-90° (from EtOH), and 0.74 g. VI. *erythro*-PhCH(*O*<sub>2</sub>*CC*<sub>6</sub>*H*<sub>5</sub>*NO*<sub>2</sub>-*p*)CH<sub>2</sub>NH<sub>2</sub>HCl, 0.74 g. VI, was obtained in a 9% yield by treating 1 min. m. 210°, was obtained in a 9% yield by treating 1 min. with ice and extg. with AcOEt; 89% of XXIV was recovered, m. 110-1°. Transformation of the 2-oxazolines to *p*-chloroacetyl amides was effected by passing dry HCl into a cooled dioxane suspension of the oxazoline, by heating the

mixt., until all of the HCl salt dissolved, and by evap't. the soln. (method A), or by removing the solvent from the HCl salt ppnd., and by heating the residual 3-6 min. at 190-92° (method B), or by using in tact hot DMSO instead of dioxane (method C). Thus were prep'd. the following *p*-chloroacetyl amides, method of prep't., yield in %, m.p. (from C<sub>6</sub>H<sub>6</sub>), and R, R', % yield and m.p. of *p*-O<sub>2</sub>NCH<sub>2</sub>CO<sub>2</sub>CHRCHR'NH<sub>2</sub>HCl obtained by refluxing the chloroacetyl amide 1.5 hrs. on the steam-bath in 75% EtOH: III, A, 60, 132-3°; Ph, Me, *erythro*, 28%; Me, *threo*, 95, 210°; V, II, 57, 128-9°; Ph, Me, *erythro*, 28%; Me, *threo*, 95, 210°; V, II, 57, 128-9°; Ph, Me, *erythro*, 28%; Me, *threo*, 62, 198°; *threo*-MeCHClCHPh-*NHCOCH\_2NO*<sub>2</sub>-*p*, B, 49, 106-7°; Me, Ph, *erythro*, 33%; *threo*-*CH\_2CHClCHMeNHCOCH\_2NO*<sub>2</sub>-*p*, C, 30, 202°; *erythro*-*CH\_2CHClCHMeNHCOCH\_2NO*<sub>2</sub>-*p*, C, 55, 190°; *threo*-*CH\_2CHClCHMeNHCOCH\_2NO*<sub>2</sub>-*p*, C, 75%, 140-1°; recovered 100% after the hydrolysis with 75% EtOH. IV. Reaction of 2-oxazolines with thiogacetic acid. New syntheses of *cis*-amino thiol. Miroslav Svoboda, Jiri Sicher, Jiri Farkas, and Magdalena Pechkovska, *Chem. Listy* 49, 1551-62; *Collection Czechoslov. Chem. Commun.* 20, 1426-38(1955)(in English).—Epimeric 2-oxazolines of the general formula RCH<sub>2</sub>CHR'N(CR")<sub>2</sub>O react with AcSH to give RCH<sub>2</sub>CHR'N(CAc)<sub>2</sub>O. The reaction is not general and fails with certain derivs. The 2-oxazolines having R' in position 5 show different reactivity for the *cis*- and *trans*-isomers, the latter being much less reactive. *cis*-2-Phenyl-4,5-trimethylene-2-oxazoline (I) (*cis*-aminoxylopeten-4-en-5-one) gave normal *S*-acetyl-*N*-benzyl-*cis*-2-aminoxylopeten-4-en-5-one (II), whereas the oxazolines derived from *cis*- and *trans*-2-phenyl-4,5-dimethylaminoxylohexanol, (*cis*- and *trans*-2-phenyl-4,5-dimethyl-*cis*-2-oxazolines (III, IV) reacted with AcSH differently and gave *N*-acetyl-*N*-thiobenzyl-*cis*- (and *trans*)-2-aminoxylohexanol (V, VI) which were transformed by means of Ar<sub>2</sub>O<sub>3</sub>

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to *cis*- and *trans*-benzoyl analogs (VII, VIII) of V and VI. The *cis* deriv. III gave the normal product *S*-acetyl-*N*-benzoyl-*cis*-2-aminocyclohexanethiol (IX) in addn. to V. The different behavior is explained by different distortion of the normal cyclic conformations by the annulation with the 5-membered oxazoline ring. This annulation makes the C—O bond almost inaccessible in the case of IV. The diacylamino thiols which could not be prep'd. by the above procedure from the oxazolines were prep'd. by the reaction of AcSH with ethyleneimines. During the prepn. of oxazolines from the amino alcohols by the reaction with Et benzimidate, the three epimers showed much smoother reaction course than the *erythro* derivs. The 2-oxazolines were prep'd. as follows: 0.01 mole of the amino alcohol was refluxed with 0.01 mole of Et benzimidate 4-8 hrs. on the steam-bath (in some cases 5-8 hrs. at 120-30°), the unreacted imidate was distd. off, and the residue was crystd. or subjected to chromatography or distn. (method A). *cis*-PhCH(OH)CHPhNBz (4.35 g.) was added to coned. H<sub>2</sub>SO<sub>4</sub> (20 ml.), the mixt. was allowed to stand 10 min. at room temp., mixed with ice, the mixt. neutralized with 30% NaOH, and the product extd. with Et<sub>2</sub>O and C<sub>6</sub>H<sub>6</sub> to give 3.02 g. of the *trans*-oxazoline (method B). The soln. of 0.01 mole amino alcohol in 10 ml. CHCl<sub>3</sub> was mixed with 0.01 mole MeC(OEt)<sub>2</sub> NH·HCl in 10 ml. CHCl<sub>3</sub>. The mixt. was heated 30 min. on the steam-bath; 40 ml. Et<sub>2</sub>O was added, the NH<sub>3</sub>Cl removed, and the soln. evapd. to dryness (method C). Thus were prep'd. the following RCH·CHR'·N:CR'·O (X), (R, R', R'', method of prepn..

yields in %, and m.p.s. given): *cis*-Ph, Me, Ph, A, 87.8, b, 151; *cis*-Ph, Ph, Ph, A, 88.1-89.8%, 83-4° and 102-3° (dimorphous) (from petr. ether); *trans*-Ph, Ph, Ph, B, 73.5, 83-5° (from petr. ether), 93-4° (from MeOH) (dimorphous); *cis*-Ph, CO<sub>2</sub>Et, Me, C, 67, b, 130-1"; *trans*-Ph, CO<sub>2</sub>Et, Me, C, 78, 132° (from EtOH); *trans*-*p*-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>CO<sub>2</sub>Et, Me, C, 67, 89-90° (from cyclohexane); *trans*-Ph, Ph, Me, A, 59, b, s, 160-1"; *cis*-C<sub>6</sub>H<sub>5</sub>, Me, Ph, A, 94, 81-3° (from ligroine); *trans*-C<sub>6</sub>H<sub>5</sub>, Me, Ph, A, 91, b, 128"; *cis*-Me, Ph, Ph, A, 90, b, 143". The oxazoline (0.01 mole) was dissolved in 2-4 ml. AcSH, the soln. was heated on the steam bath for the time given, the AcSH was evapd. in vacuo, C<sub>6</sub>H<sub>6</sub> was added and again evapd. [R, R', and R'' in X, time (in hrs.) and temp. of heating; yield in %, and m.p. of RCH(SAc)CHR'NHCOR' are given]: *cis*-Ph, Me, Ph, 5, 100°, 76%, 133-5° (from iso-PrOH); *trans*-Ph, Me, Ph, 4, 100°, 72%, 128-0.5° (from iso-PrOH); same, 24 hrs. at 20°, 79%; *cis*-Ph, Ph, Ph, 4, 100°, 38%, 152° (from iso-PrOH or C<sub>6</sub>H<sub>6</sub>); *trans*-Ph, Ph, Ph, 2, 100%, 95%, 225° (from C<sub>6</sub>H<sub>6</sub>); same, 12 hrs., 20°, 71%; *trans*-Ph, CO<sub>2</sub>Et, Ph, 4, 100°, 72%, 96-8° (from iso-PrOH); same, CHCl<sub>3</sub>-COSH, 2, 100%, PhCH(SCOCHCl<sub>3</sub>)CH(CO<sub>2</sub>Et)NHCOPh; *trans*-Ph, CO<sub>2</sub>Et, Me, 0.5, 100°, 92%, 97-8° (from C<sub>6</sub>H<sub>6</sub>-petr. ether); *trans*-*p*-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CO<sub>2</sub>Et, Me, 1, 100°, 73%, 139-40° (from C<sub>6</sub>H<sub>6</sub>-petr. ether); *trans*-Ph, Ph, Me, 4, 100°, 41%, 167-8°; *cis*-C<sub>6</sub>H<sub>5</sub>, Me, Ph, 5, 100°, 33%, 125° (from petr. ether); *cis*-Me, Ph, Ph, 5, 100°, 54%, 120-7° (from iso-PrOH). Refluxing 9 g. I with 9 ml.

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AcSH 1 hr. on the steam-bath gave 7.9 g. II, m. 136-7° (from C<sub>6</sub>H<sub>6</sub>). Refluxing *trans*-N-benzoyl-S-acetylaminocyclopentanethiol (1.32 g.) in 50 ml. EtOH 3 hrs., with 3 g. Al(iso-PrO)<sub>3</sub>, treating the mixt. with 200 ml. 5% tartaric acid, and extg. the product with AcOEt gave 0.77 g. *trans*-N-benzoyl-2-amino-cyclopentanethiol, m. 123-8°. Heating 9.6 g. IV with 9.6 ml. AcSH 2 hrs. on the steam-bath yielded 8.85 g. VI, m. 159-60°. Heating 1.5 g. VI in a mixt. of dioxane and water 1:1 with 4.5 g. Ag<sub>2</sub>O, 1.5 hrs. on the steam-bath, filtering the mixt., and evapg. the filtrate gave VIII, m. 120-1° (from ligroine). Heating 8.93 g. III with 9 ml. AcSH 2 hrs. on the steam-bath, distg. off the AcSH, adding petr. ether, and boiling the product with Et<sub>2</sub>O gave sol. V, yellow needles, m. 139-40° (from C<sub>6</sub>H<sub>6</sub>-petr. ether), and 1.73 g. sol. *trans*-N-benzoyl-S-acetyl-2-amino-cyclohexane-thiol, m. 146° (from iso-PrOH). *trans*-N-Acetyl-2-amino-cyclohexanol (XI), m. 124-5° (from AcOEt) was prep'd. by passing 10 min. CH<sub>2</sub>:CO into a soln. of 1.15 g. *trans*-2-amino-cyclohexanol in 30 ml. H<sub>2</sub>O, by allowing the mixt. to stand 3 hrs., by evapg. the mixt., and crystg. the residue from AcOEt. XI and BzCl gave *trans*-N-acetyl-O-benzoyl-2-amino-cyclohexanol, m. 120-1° (contrary to the literature). Adding 1.2 ml. BzCl in 10 ml. Et<sub>2</sub>O to a cooled soln. of 1 g. cyclohexenimine (XII) in 20 ml. Et<sub>2</sub>O and 25 ml. 20% NaOH, sepg. the Et<sub>2</sub>O layer, and evapg. the solvent gave 1.5 g. N-benzoylcyclohexenimine (XIII), m. 77-8° (from C<sub>6</sub>H<sub>6</sub>-petr. ether). Mixing 0.3 g. XIII with 0.6 ml. AcSH gave 0.34 g. *trans*-N-benzoyl-S-acetyl-2-amino-cyclohexanethiol, m. 144-5° (from C<sub>6</sub>H<sub>6</sub>-petr. ether). Treating a soln. of 1 g. XII in 3 ml. C<sub>6</sub>H<sub>6</sub> with 2.4 ml. AcSH in 3 ml. C<sub>6</sub>H<sub>6</sub> and evapg. the solvent gave 0.3 g. *trans*-N-acetyl-2-amino-cyclohexanethiol, m. 120-2° (from aq. EtOH). Heating a soln. of 1.8 g. *cis*-2,3-diphenylethlenimine in 3 ml. AcSH 4 hrs. on the steam-bath, distg. the excess AcSH *in vacuo*, and crystg. the residue from aq. EtOH gave 0.9 g. *trans*-PhCH(S-

(S*t*)CHPhNHAc, 140-1°. V. Synthesis and configuration of both racemic  $\beta$ -phenylcysteines. Jiri Sicher, Miroslav Svoloda, and Jiri Farkas. *Chem. Listy* 49, 1363-74; *Collection Czechoslov. Chem. Commun.* 20, 1429-51 (1955) (in English).—Contrary to AcSH, CHCl<sub>2</sub>COSH adds to PhCH(C(CO<sub>2</sub>Et)NHbz) (I) and  $\rho$ -O<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>C(CO<sub>2</sub>Et)NHbz (II) and gives *threo*-PhCH(SCOCHCl<sub>2</sub>)CH(CO<sub>2</sub>Et)-NHbz (III) and *threo*- $\rho$ -O<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>SCOCHCl<sub>2</sub>)CH(CO<sub>2</sub>Et)NHbz (IV), resp. Hydrolysis of III gave *threo*-PhCH-(SH)CH(CO<sub>2</sub>Et)NHbz (V), the acetylation of which yielded *threo*-PhCH(SAc)CH(CO<sub>2</sub>Et)NHbz (VI). Acidic hydrolysis of V and its *erythro*-epimer, *erythro*-PhCH(SH)CH<sub>2</sub>(CO<sub>2</sub>Et)NHbz (VII) (see part II above), give *trans*-2,5-diphenyl-4-carboxy-2-thiazoline (VIII) and HCl salt of *threo*-PhCH(SH)CH(CO<sub>2</sub>H)NH<sub>2</sub> (IX). The same result was obtained by the hydrolysis of *cis*- and *trans*-ethyl ester (X, XI) of VIII. VIII was also obtained from PhCH(OEt)-NH and HCl salts of IX and *erythro*-PhCH(SH)CH<sub>2</sub>(CO<sub>2</sub>H)NH<sub>2</sub> (XII). The action of HCl in EtOH transformed *erythro*-PhCH(SAc)CH(CO<sub>2</sub>Et)NHAc (XIII) to *erythro*-PhCH(SH)CH(CO<sub>2</sub>Et)NH<sub>2</sub>·HCl (XIV) and to 2-methyl-4-carbethoxy-5-phenyl-2-thiazoline (XV). Hydrolysis of XIV, XII·HCl, or XV gave IX·HCl. XIII is thus a convenient material for the prep'n. of both racemic  $\beta$ -phenylcysteines IX and XII. Heating 13.3 g. 2-phenyl-5-p-nitrobenzylideneoxazolidone in 100 ml. EtOH and 5 ml. concd. H<sub>2</sub>SO<sub>4</sub> 2 hrs. on the steam-bath gave 12.3 g. II, m. 165-6°. Heating a mixt. of CHCl<sub>2</sub>COSH (25 ml.) with 34 g. I 5 hrs. on the steam-bath, adding Et<sub>2</sub>O after cooling, and crystg. the white crystals from EtOH gave 24.3 g. III, m. 123-31°. Similarly heated 37.5 g. IV, m. 142-3°. Refluxing 2.2 g. III with 1 g. (iso-PrO)<sub>2</sub>Al in 50 ml. EtOH 4 hrs., evapg. the soln. to dryness, treating the residue with 15% tartaric acid, and extg. the cryst. product with Et<sub>2</sub>O gave 1.2 g. V, m. 125-6° (fresh iso-PrOH). Similarly

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*erythro-PhCH(OCOCH<sub>2</sub>)CH(CO<sub>2</sub>Et)NHBz* gave **VII**, m. 117-10°, quantitatively. **V** and Ac<sub>2</sub>O in pyridine gave *PhCH(SAc)CH(CO<sub>2</sub>Et)NHBz*, m. 126-8° (from iso-PrOH). Heating 1.55 g. **X** (for the prepn. see part II) with 20 ml. EtOH satd. with HCl 20 hrs. on the steam-bath, evapg. the soln. to dryness, dissolving the residue in H<sub>2</sub>O, and extg. with Et<sub>2</sub>O gave 1.28 g. of an oil which was sepd. by extn. with petr. ether into fusol. **VIII**, m. 137-8° (from C<sub>6</sub>H<sub>6</sub>), and 350 mg. sol. **XI**, m. 65-6° (from petr. ether). Treating 620 mg. **X** (or **XI**) in 1 ml. EtOH with 150 mg. NaOH in 2 ml. EtOH, acidifying the soln. with HCl and extg. with Et<sub>2</sub>O gave 620 mg. **VIII**, erroneously formulated as *PhCH(C(CO<sub>2</sub>H)NC<sub>2</sub>H<sub>5</sub>)Ph* by Lurje and Gacenko (*C.A.* 47,

2168c); *Na salt* of **VIII**, m. 262°, obtained by NaOH in EtOH from **X** and **XI**; *picrate* of **VIII**, m. 165-7° (from iso-PrOH); *HCl salt* of **VIII**, m. 165-7° (from iso-PrOH-Et<sub>2</sub>O). Adding Ph(C(OEt)<sub>2</sub>)NH (0.33 g.) to a soln. of 0.47 g. HCl salt of **IX** under N, dig. the mixt. after 24 hrs. with H<sub>2</sub>O, evapg. *in vacuo*, dissolving the residue in Et<sub>2</sub>O-C<sub>6</sub>H<sub>6</sub> mixt. 1:1, washing the soln. with 1% HCl, evapg. the soln., and crystg. the residue from C<sub>6</sub>H<sub>6</sub> gave **VIII**. The same compnd. was obtained from **XII**. Heating 1.65 g. **V** with 35 ml. 10% HCl 5 hrs. on the steam-bath and extg. the mixt. with Et<sub>2</sub>O gave 0.35 g. **VIII**, and from the aq. layer 0.17 g. **IX.HCl**, m. 198-9° (from EtOH-petr. ether). Similarly **VII** gave 38% **VIII** and **IX.HCl**. Heating 28.33 g. **VIII** 10 hrs. on the steam-bath with 700 ml. 10% HCl gave 0.5 g. **VIII.HCl**-by ether extn. BzOH (6.9 g.), 2.1 g. recovered **VIII**, and by evapn. of the aq. layer 7.1 g. of **IX.HCl**, m. 198-9°, which was transformed by esterification and benzylation to *threo-PhCH(SBz)CH(CO<sub>2</sub>Et)NHBz*, m. 181°. Heating a suspension of 13.25 g. *PhCH(C(CO<sub>2</sub>H)NC<sub>2</sub>H<sub>5</sub>)Ph* in 200 ml. EtOH

satd. with HCl 4 hrs. on the steam-bath, evapg. the soln. to dryness *in vacuo*, adding H<sub>2</sub>O, extg. the mixt. with Et<sub>2</sub>O, evapg. the ext. to dryness, and extg. the residue with petr. ether gave 400 mg. insol. **VIII**, 3.0 g. sol. **X**, m. 93-4°, and 3.7 g. **XI**. Refluxing 48 g. **XIII** with 200 ml. EtOH (satd. with HCl) 10 hrs. on the steam-bath give 15.1 g. **XIV**, m. 180-1° (from iso-PrOH-Et<sub>2</sub>O), and from the mother liquors by extn. with Et<sub>2</sub>O, 8.82 g. **XV**, b.p. 125-33°. Heating 7.80 g. **XIV** with 35 ml. 10% HCl 4 hrs. on the steam-bath gave 4.05 g. of the **XII.HCl**, m. 90-100° solidifying and m. again at 178-80° (from H<sub>2</sub>O), 184-5° (from iso-PrOH-petr. ether). Heating 2.6 g. **XV** with 14 ml. 10% HCl 5 hrs. on the steam-bath gave 1 g. **IX.HCl**, m. 204-5° (from EtOH-Et<sub>2</sub>O). Similar hydrolysis of 1.55 g. **XIII** yielded 220 and 60 mg. of the HCl salts of **IX** and **XI**, resp. Heating 1.65 g. **XIII** with 20 ml. N NaOH 30 min. on the steam-bath under N, acidifying the soln. with HCl to pH 2, and crystg. the crystals from H<sub>2</sub>O gave 0.68 g. *erythro-PhCH(SH)CH(CO<sub>2</sub>H)NHAc*, m. 160°. Heating 1.55 g. **XIII** with 1.4 g. Ba(OH)<sub>2</sub> in 20 ml. H<sub>2</sub>O 10 hrs., removing Ba with H<sub>2</sub>SO<sub>4</sub>, and evapg. the filtrate yielded 800 mg. *threo-PhCH(SH)CH(CO<sub>2</sub>H)NHAc*, m. 146° (from H<sub>2</sub>O). Esterification of the HCl salts of **IX** and **XI** with HCl in EtOH by refluxing 1 and 6 hrs., resp., gave 43% *threo-epimiz* of **XIV**, m. 148-9° (from EtOH-petr. ether), and **XIV**, resp. **IX.HCl** was esterified and acetylated with Ac<sub>2</sub>O in pyridine to give *threo-PhCH(SAc)CH(CO<sub>2</sub>Et)NHAc*, m. 105.5° (from C<sub>6</sub>H<sub>6</sub>-petr. ether). Similar treatment of **XIV** gave 55.7% **XII**, m. 90° (from C<sub>6</sub>H<sub>6</sub>-petr. ether). *threo-PhCH(SH)CH(CO<sub>2</sub>Et)NHBz* and **XIV** gave *threo*- and *erythro-PhCH(SBz)<sub>2</sub>CH(CO<sub>2</sub>Et)NHBz*, m. 180-2° and 160-1°, resp. VI. Configuration of 1-phenyl-1,2-propanediols. Steric course of the hydrolysis of *cis*- and *trans*-1-phenyl-2-methylethylene oxide. Miroslav Sloboda and Ilii Sicher. *Chem. Listy* 49, 1375-8;

**JIRI SICHER**

*Collection Czechoslov. Chem. Commun.*, 20, 1432-5 (1955) (in English).—The reaction of  $\text{HCO}_2\text{H}$  with *cis*- (I) and *trans*- $\text{PhCH}(\text{CHMe})_2$  (II) as well as the acid-catalyzed fission of *cis*- (III) and *trans*- $\beta$ -methylstyrene oxide (IV) give a mixt. of epimers and are not, contrary to the general opinion, sterically homogeneous. The  $\text{HCO}_2\text{H}$  oxidation of I produced a mixt. with prevailing *cis*- $\text{PhCH}(\text{OH})\text{CHMeOH}$  (V) whereas I yielded a mixt. with prevailing *trans*- $\text{PhCH}(\text{OH})\text{CHMeOH}$  (VI). Similar results were obtained by the fission of III and IV. Pure V and VI were prepd. by the oxidation with  $\text{KMnO}_4$  of I and II, resp. The higher melting  $\beta$ -isomer (m. 91-2°) was found to be V, the lower melting  $\alpha$ -isomer (m. 54-6°) was VI. Adding 5.2 g. II to a mixt. of 31.5 g.  $\text{HCO}_2\text{H}$  and 6.05 ml. 38.5%  $\text{H}_2\text{O}_2$ , stirring the mixt. 45 min. at 40°, allowing the mixt. to stand 18 hrs. at room temp., distg. off the excess  $\text{H}_2\text{O}_2$  and  $\text{HCO}_2\text{H}$  *in vacuo*, drying the residue (7.7 g.) by distn. with  $\text{CaH}_2$ , dissolving the residue in 60 ml.  $\text{Et}_2\text{O}$ , treating the soln. with 56 ml. of an ether soln. contg. 31 mg.  $\text{LiAlH}_4$  per ml., heating the mixt. 1 hr. on the steam-bath, decoupg. with 4.9 ml.  $\text{H}_2\text{SO}_4$  in 90 ml.  $\text{H}_2\text{O}$ , extg. with  $\text{Et}_2\text{O}$ , and chromatographing the residue after the evapn. of the  $\text{Et}_2\text{O}$  gave by elution 1.3 g. of an oil not further identified, by elution with  $\text{Et}_2\text{O}$  0.25 g. V, m. 85-8°, and by  $\text{EtOH}$  eluted 1.27 g. VI, m. 50-52°. Similar results were obtained by hydrolyzing the intermediate half-formate with  $\text{NaOH}$ . I (b. 31-2°, m. 1.5399) (5.9 g.) and a mixt. of 31.5 g.  $\text{HCO}_2\text{H}$  and 6.05 ml. 38.5%  $\text{H}_2\text{O}_2$  gave in the above manner a small amt. of impure V and 2.45 g. crude VI, m. 38-42°. III was obtained from

perphthalic acid and II in a 73.5% yield, b. p. 78-80°, and IV from perphthalic acid and II in a 79.8% yield, b. p. 93-5°, m. 1.5143. Refluxing a mixt. of 1 g. II, 6 ml.  $\text{H}_2\text{O}_2$ , 5 ml. dioxane, and 0.1 ml.  $\text{H}_2\text{SO}_4$  3 hrs., neutralizing it with  $\text{NaHCO}_3$ , evapg. *in vacuo*, extg. the residue with  $\text{Et}_2\text{O}$ , evapg. the  $\text{Et}_2\text{O}$  and chromatographing the residue gave 960 mg. of a semicryst. residue, b. p. 102-6°, yielding 800 mg. of a mixt. contg. 63% V and 37% VI (according to the effect of both isomers on the titration const. of phenylboric acid). Similarly 1 g. III gave 760 mg. of a mixt. b. p. 23-3°, contg. 20% V and 80% VI. A soln. of 5.2 g. II in 60 ml.  $\text{EtOH}$  was treated at -40° with a soln. of 10.4 g.  $\text{KMnO}_4$  and 10 g.  $\text{KgSO}_4 \cdot 7\text{H}_2\text{O}$  in 350 ml.  $\text{H}_2\text{O}$ , the  $\text{MnO}_4^-$  was removed, washed with hot  $\text{EtOH}$ , and the filtrate was evapd. *in vacuo*. The cryst. residue was dissolved in a min. amt.  $\text{H}_2\text{O}$ , extd. with  $\text{Et}_2\text{O}$ , the  $\text{Et}_2\text{O}$  was evapd., and the residue (2.5 g.) crystallized from  $\text{Et}_2\text{O}$ -petr. ether to give V. Similarly II gave 51% VI.

J. Hradilek

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*Sicher*

*Chem*

Polarographic determination of rate of periodate oxidation of epimeric open-chain 1,2-glycols. / P. Zuman, J. Sicher, J. Krupička, and M. Svoboda (Czech. Acad. Sci., Prague). Nature 178, 1407-8 (1959). The rates of fission of a series of stereoisomeric pairs of open-chain dissecondary glycols by periodic acid are detd. polarographically. The specific 2nd-order rate consts. ( $1.$  mole $^{-1}$  sec. $^{-1}$ ) for the isomeric pairs (given in the order *threo, erythro*) in  $6 \times 10^{-4} M$   $KIO_4$  and  $8 \times 10^{-4} M$  glycol aq. soln. at  $25^\circ$  are as follows: 1-phenylpropane-1,2-diol, 9.2, 3.7; butane-2,3-diol, 9.9, 4.8; hexane-3,4-diol, 8.8, 4.4; octane-4,5-diol, 5.5, 4.0; 1-cyclohexylpropane-1,2-diol, 5.6, 4.6; 1-phenylpropane-1,2-diol, 10.1, 4.6; 1-(*p*-methoxyphenyl)propane-1,2-diol, 9.8, 5.2; 1-phenylbutane-1,2-diol, 8.2, 4.9; 1-phenyl-3-methylbutane-1,2-diol, 11.5, 3.7; 1,2-diphenylethane-1,2-diol, 14.2, 3.3. The effect of acid and base catalysis (using  $HNO_3$ ,  $HOAc$ ,  $NaOAc$ ,  $NaH_2PO_4$ , and  $Na_2HPO_4$ ) on the rate of reaction for the two pairs is also given. The effects of mol. structure on the rate consts. are discussed.

R. Benz

Country : Czechoslovakia G-1  
Category : Organic Chemistry, Theoretical Organic  
Chemistry  
Aba. Jour. : Ref. Zhur.-Khimiya No. 6, 1959 19286  
Author : Zuman, F.; Sicher, J.; Krupicka, J.; Svecboda, M.  
Institut. :  
Title : Stereochemical Studies. VII. Periodate Oxidation  
of Diastereoisomeric Diols of the Type R.CH(OH).  
CH(OH).R'.  
Orig Pub. : Collect. czechosl. chem. commun., 1958, 23,  
No 7, 1237-1251  
Abstract : See RZhKhim, 1958, 43207.

Card: 1/1

N-tritylserine (II), N-tritylserine (III), 1-trityl-2-carbomethoxyethylene-  
imine (IV), which on reaction with NH<sub>2</sub>OH.HCl gives

APPROVED FOR RELEASE: 03/14/2001 CIA-RDP86-00513R001550420009-5"  
Card : 1/10

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances  
and their Synthetic Analogs. G-3

Abs. Jour: Referat Zhur-Khimiya, No 4, 1953, 11467.

hydroamic acid (V); the latter adds HCl with the formation of the hydrochloride of  $\alpha$ -amino- $\beta$ -chloropropiohydroxamic acid (VI); strongly basic anion exchange resins cyclize VI to I. For comparison purposes 4-benzylamine- (IX) and 4-benzhydrylaminooisoxazolidone-3 (X) were synthesized from N-benzyl-2-carbomethoxyethyleneimine (VII) and N-benzhydryl-2-carbomethoxyethyleneimine (VIII) by the same method. 1-benzylethylenimine-2-carbhydroxamic acid (XI) is synthesized by refluxing 87.5 gms of the methyl ester of 1,2-dibromopropionic acid [sic] in 550 ml C<sub>6</sub>H<sub>6</sub> for 3 hrs with 71.4 gms triethylamine and 38.2 gms benzylamine, shaking the mixture with water for 12 hrs and allowing the mixture to stand with VII [TN: meaning appears garbled] for 48 hrs, obtained by eva-

Card : 2/10

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001550420009-5

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances  
and their Synthetic Analogs. G-3

Abs. Jour: Referat Zhur-Khimiiya, No 4, 1958, 11467.

porating the benzene solution with cold ( $5^{\circ}$ ) hydroxyl-amine (prepared from 49.6 gms of the hydrochloride of hydroxylamine in 250 ml  $\text{CH}_3\text{OH}$  and 24.3 gms Na in 300 ml  $\text{CH}_3\text{OH}$ ) in 50 ml  $\text{CH}_3\text{OH}$ , followed by evaporation to 200 ml at  $20^{\circ}$ . XI is isolated by dilution with water and neutralization with  $\text{CH}_3\text{COOH}$ , yield 73.6%, mp  $154$ - $155^{\circ}$  (from 99% alcohol). For proof of structure 0.3 gm XI is hydrogenated over 0.2 gm  $\text{PtO}_2$  in 10 ml  $\text{CH}_3\text{COOH}$  and the product is refluxed for 3 hrs with 5 ml (1 : 1) HCl (acid); paper chromatography using the system phenol-water- $\text{NH}_3$  has established the presence of alanine (XII) and  $\alpha$ -alanine (7 : 3) in the reaction mixture. Dry HCl gas is passed for 30 min at  $0^{\circ}$  into 20 gms I in 200 ml benzene, followed

Card : 3 / 10

CZECHOSLOVAKIA/Organic Chemistry: Naturally Occuring Substances  
and their Synthetic Analogs. G-3

Abs. Jour: Referat Zhur-Khimiya, No 4, 1958, 11467.

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances  
and their Synthetic Analogs.

G-3

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11467.

0.5 liter  $\text{CH}_3\text{OH}$  with 12.5 gms  $\text{Na}_2\text{CO}_3$  in 1 liter water, allowing to stand 4 days, and acidifying with 50 ml  $\text{CH}_3\text{COOH}$  in 200 ml alcohol; the yield is 73%, mp 139-141° (decomp: from  $\text{CH}_3\text{OH}$ ). Hydrogenation and hydrolysis of the latter product give serine. The hydrogenation of 0.03 mol X in 100 ml alcohol and 1 ml  $\text{CH}_3\text{COCH}_3$  over  $\text{PtO}_2$  by 200 ml  $\text{H}_2$  gives 2.15 gms of the amide of N-benzhydrylserine, mp 142-144° (from alcohol). A mixture of 0.03 mol II, 50 ml dry pyridine, and 2.5 ml mesyl chloride is kept 24 hrs at 0.4°, diluted with 200 ml water and  $\text{CHCl}_3$ ; III is extracted in 90% yield, mp 128° (from benzene-alcohol) 0.01 mol II in 50 ml dioxane is mixed with  $\text{NH}_2\text{OH}$  (prepared from 14 gms  $\text{NH}_2\text{OH}\cdot\text{HCl}$  in 100 ml abs

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CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances and their Synthetic Analogs.

G-3

Abs Jour: Referat Zhur-Khimiya, No 4, 1958, 11467.

$\text{CH}_3\text{OH}$  and 6.9 gms Na in 30 ml  $\text{CH}_3\text{OH}$ ); after 3 days the reaction mixture is diluted with 100 ml water, and neutralized with  $\text{CH}_3\text{COOH}$ ; further dilution results in the formation of a precipitate (33 gms) of N-tritylserinehydroxamic acid, mp 109° ( $\text{CH}_3\text{OH}$ ); the product contains 1 molecule of combined  $\text{CH}_3\text{OH}$ . A mixture of 0.02 mol III, 20 ml  $\text{C}_6\text{H}_6$ , 1 ml  $\text{CH}_3\text{CH}_2$ , and 2.5 gms N-ethylpiperidine is refluxed 8 hrs, diluted with 15 ml  $\text{CHCl}_3$ , washed with water, and evaporated; the yield of IV is 80%, mp 130-131° (from benzene-cyclohexane). On standing for 3 days a mixture of 0.18 mol IV in 100 ml dioxane and 25.2 gms  $\text{NH}_2\text{OH}\cdot\text{HCl}$  and 12.4 gms Na in 150 ml  $\text{CH}_3\text{OH}$

Card : 8/10

CZECHOSLOVAKIA/Organic Chemistry. Naturally Occurring Substances and their Synthetic Analogs.

G-3

CZECHOSLOVAKIA/Organic Chemistry. Theoretical and General  
Questions on Organic Chemistry.

C-1

Abs Jour: Ref Zhur-Khim., No 13, 1958, 43207.

Author : Zuman Petr, Sicher Jiri, Krupicka Josef, Svoboda  
Miroslav.

Inst :

Title : Stereochemical Studies. VII. Oxidation of Diastereo-  
isomeric Diols of RCH(OH)CH(OH)R' Type with Periodate.

Orig Pub: Chem. listy, 1957, 51, No 6, 1068-1081.

Abstract: Polarographic study (see Communication VI, RZhKhim,  
1956, 78180) of the rate of oxidation of nine pairs  
of acyclic diols of RCH(OH)CHCHR' type with periodate  
at different pH (2-7.9) and diol concentration  
( $6 \cdot 10^{-5}$  -  $9 \cdot 10^{-4}$  M). Investigated were  
ethylene glycol (I), threo- and erythro-isomers of

Card : 1/3

1134 CZECHOSLOVAKIA  
AVAILABILITY: Scientific literature, 1958-1961  
A.D. JOUR.: RUMHA., No.10, 1958, p. 2212  
AUTHOR: Svoboda, M.; Tichy, L.; Sicher, J.  
TITLE: Stereochemical Studies. XI. Synthesis of cis-  
and trans-2-amino cyclohexadecanol and 1-amino-  
cyclopentadecanol  
SHP. PUB.: Collect. czechoslov. chem. commun., 1958, 23,  
no 10, 1958-1961; Chem. Listy, 1958, 52, 1961  
ABSTRACT: In order to study the relation between the  
configuration and chemical and physical prop-  
erties, cis- and trans-2-amino cyclohexadecanol (cis-  
and trans-I) and cis- and trans-  
2-amino cyclopentadecanol (cis- and trans-II)  
were synthesized. Through the hydrogenation  
of the oxime of 2-oxy cyclohexadecanone (III)  
over Pt (from PtO<sub>2</sub>) in alcohol at ~30° and  
normal pressure, cis-I was obtained, yielding  
21.65 g (from 100 g III), m.p. 120-121° (from

CD:

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3-6

COUNTRY :  
CAT. NO. :  
AEC. JOUR. : REKHAIA., No. 23 1952, No. 22241

AUTHOR :  
INST. :  
TITLE :

ORIG. PUB. :

ABSTRACT cont'd. : yield 90%, m.p. 142-142.5° (from acetone); NBD, m.p. 162-163.5° (from alcohol). By boiling (18 hours) trans-2-phenyl-4,5-dodecamethylene- $\Delta^2$ -oxazoline with concentrated HCl in alcohol, trans- $\text{I}$  was obtained, yielding 93.5%, m.p. 105-106° (from petroleum ether); HC, m.p. 215-216° (from alcohol-ether); ED, yield 88.5%, m.p. 170-171° (from alcohol); NBD, m.p. 192-193° (from alcohol). Analogously, from trans-2-phenyl-4,5-tridecamethylene-

CARD: 3/3

G-7

COUNTRY :  
CATALOGUE :  
ARTS. JCUR. : RZKhim., No. 23 1959, No. 62261

AUTHOR :  
TITLE :  
SERIAL :

ORIG. PUP. :

ABSTRACT  
cont'd : cis-I, 99.5, 148-149; trans-I, 97, 129-130;  
cis-II, 100, 120-121; trans-II, 95, 107-108.  
The obtained methane sulfonates being heated  
with  $\text{CH}_3\text{COOK}$  in alcohol (10-30 hours at 30-  
 $95^{\circ}\text{C}$ ) are transformed into 2-phenyl-4,5-poly-  
methylene- $\Delta^2$ -oxazoline, the configuration of  
which is opposite to the initial amino-alco-  
hols [the initial amino-alcohol, polymethyl-  
ene, yield in %, m.p. in  $^{\circ}\text{C}$  (from petroleum  
ether), b.p. in  $^{\circ}\text{C}/\text{mm}$ , m.p. of picrate in  $^{\circ}\text{C}$ ]

SEARCHED: 5/8

G-8

COUNTRY :  
CATEGORY :

ASS. JOUR. : AZKhim., no. 23 1959, No. 522H1

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ORIG. PUB. :

**ABSTRACT** : are given]: cis-I, trans-dodecamethylene, 91.6, 61-63, 190-200/0.2-0.3, 157-160 (from CH<sub>2</sub>Cl<sub>2</sub>, 61-63, 190-200/0.2-0.3, 157-160); trans-I, cis-dodecamethylene, 71, alcohol); trans-I, cis-dodecamethylene, 71, alcohol); cis-II, 95-96, --, 160-161 (from alcohol); cis-II, 95-96, --, 160-161 (from alcohol); cis-III, 100-103-tridecamethylene, 91.6, 67-68, 160-161 (from CH<sub>2</sub>Cl<sub>2</sub>); trans-III, 100-103, 160-161 (from CH<sub>2</sub>Cl<sub>2</sub>); cis-IV, 91.6, 67-69, 120-122-tridecamethylene, 91.6, 67-69, 120-122/0.2, 130.5-137 (from CH<sub>2</sub>Cl<sub>2</sub>). In IR and NMR spectra of trans-I and trans-III left standing in LSC of trans-I and trans-III left standing in cyclohexane, saturated with gaseous HCl, cyclohexane, saturated with gaseous HCl, cyclo-

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COUNTRY :  
CATEGORY :

ASS. JOUR. : RZKlim., No. 23 1959, No. 82241

AUTHOR :  
INER. :  
TITLE :

ORIG. PUB. :

ABSTRACT cont'd : Migrates and IR of trans-2-benzoyloxy- or  
2-p-nitrobenzoyloxy-cycloalkylamines are  
formed (acyl, cycloalkyl and m.p. in °C are  
given): benzoyl, tetradecyl, 172-173; p-nitro-  
benzoyl, tetradecyl, 173-175; benzoyl, penta-  
decyl, 166.5-167; p-nitrobenzoyl, pentadecyl,  
179-181. For cis-isomers such like migration  
is not observed. The configuration of the  
products was determined according to the  
various ability of acyls for migration,

CLASS:

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3-9

SCHER, J.; TROY, V.

"Stereoechemical studies. XIII. Absolute configuration of conhydrine and  $\beta$ -conhydrine." (In English)

SECTION OF CHEMICALS AND DRUGS. Praha, Czechoslovakia,  
Vol. 23, no. 11, Nov. 1958

Monthly list of first 5000 admissions (acute). D, Vol. 1, No. 7, July 1959, Unclass.

Chem. Z., 1959, 23, 11.

"Stereoelectrostatic effect. Part II. Effect of ring size on configuration of the methanesulfonates." (In English)

Chem. Z., No. P. Chem.-Techn. Obzor vych. chem. a techn., Praha, Czechoslovakia,  
Vol. 23, no. 11, Nov. 1959

"Monthly list of new literature: Chem. Abstr. (CASA), 10, Vol. 8, No. 7, July 1959, Unclass.

STOHL, J. .. 1959, 12, 121.

"Stereocchemical studies. XIV. Effect of ring size and configuration on the rate of intramolecular cyclization of the N-thiocbenzoyl derivatives of cis- and trans-2-aminocyclanols." (In English)

CYCLIZATION OF CZECHOSLOVAK CHEMICAL COMMUNICATIONS, Praha, Czechoslovakia,  
Vol. 23, no. 12, Dec. 1959

Monthly list of EPO European Accesions (PAT), LC, Vol. 2, No. 7, July 1959, Unclass.

COUNTRY : Czechoslovakia G-2  
 CATEGORY :

ABSTRACT JOUR. : RZhKhim., No. 10 1959, No. 71412

AUTHOR : Michal Jiri; Svoboda Miroslav  
 IZM. : Not given  
 TITLE : Stereochemical Studies VIII. Synthesis of Large and Medium Ring Cis- and Trans-Aminocyclophanes.

ORIG. PUB. : Chem. Listy, 1958, 52, #6, 140-1585.  
 Coll. ct. czechoslov. chem. commun., 1958, 23, #7, 1952-1959.

ABSTRACT : The reduction of 7-12, 14 and 20 numbered rings of cyclic acyclic oximes over Pt (from PtCl<sub>2</sub>) gives a high yield of a mixture of cis and trans-2-aminocyclophanes (cis- and trans-ACO). Upon the reduction of the same oximes over Na in n-C<sub>4</sub>H<sub>9</sub>OH the basic products are aminocyclanes, while only small amounts of ACO are formed. Cis- and trans-ACO may be separated or changed into one another by means of the following procedures: A: Recrystallization of the mixture yields the less

CARD: 1/23

COUNTRY : Czechoslovakia G-2  
 CAT. NO.:  
**APPROVED FOR RELEASE: 03/14/2001 CIA-RDP86-00513R001550420009-5**  
 ABSTRACT JOUR. : RZhKhim., No. 10 1959, No. 71412

AUTHOR :  
 IZM. :  
 TITLE :  
 ORIG. PUB. :  
 ABSTRACT : soluble cis-ACO; B: Benzoylation of the residue left after recrystallization gives a mixture of derivatives of cis-and trans-ACO, enriched in trans isomer. Reaction of HCl with the above mixture causes migration of C<sub>6</sub>H<sub>5</sub>CO group, and leads largely to the formation of trans-0-benzoyl-ACO. Hydrolysis of the latter compound yields trans-ACO. C: Cis-ACO may be transformed into trans-ACO by reacting cis-N-benzoyl ACO with CH<sub>3</sub>SO<sub>2</sub>Cl, with a subsequent cyclization, in the presence of CH<sub>3</sub>COOK, of the cis-N-

CARD: 2A3

COUNTRY : Czechoslovakia  
CATEGORY :  
ABS. JOUR. : RZKhim., No. 90 1950, No. 71412  
AUTHOR :  
EDITOR :  
TRANSLATOR :  
ORIG. PUB. :  
ABSTRACT : C<sub>6</sub>H<sub>5</sub>COCl. Trans-N-benzoyl-ACO was obtained by a partial hydrolysis of the corresponding picrate (PK) of trans-FO. Following are characterized ACO's from which a N-benzoyl derivative was obtained and m.p., °C of benzoyl derivative:  
cis-II, 124 (from alc.-petr. eth.); trans-II,  
170-172 (from alc.);  
cis-III, 121-122 (from aq. alc.); trans-III,  
124-125 (from aq. alc.);  
cis-IV, 115-116 (from aq. alc.); trans-IV,  
136-139 (from benzene);  
cis-V, 152-153 (from aq. alc.); trans-V,  
129-130 (from benzene);  
cis-VI, 141-142 (from aq. alc.); trans-VI,  
119-120 (from alc.);  
7/23

SEARCHED : FILED  
SERIAL NO. :  
JULY 1969, No. 10 1969, No. 71412  
AUTHOR :  
INST. :  
DATE :  
CLIP. P.P. :  
ABSTRACT : cis-VII, 192-183 (from aq. alc.); trans-VII,  
179-180 (from alc.);  
cis-VIII, 146-147 (from aq. alc.); trans-VIII,  
141-142 (from aq. alc.);  
cis-IX, 134-139 (from alc.); trans-IX, 142-143  
(from aq. alc.);  
cis-X, 107-108 (from alc.); trans-X, 102-103  
(from alc.);  
Interaction of  $\text{CH}_3\text{COCl}$  with trans-N-benzyl-II  
(I) yields  $\beta$ -acetate, m.p. 138-141.5°(from  
benzene).  
CAND: 8/23

COUNTRY	:	Czechoslovakia	G-2
CATEGORY	:		
ABST. JOUR.	:	RZKhia., No. 20 1959, No. 71412	
AUTHOR	:		
TYPE	:		
TITLE	:		
ORIG. PUB.	:		
ABSTRACT	:	When ACC was reacted with $p$ -NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> COCl, the corresponding N-substituted ACC was obtained. Listed below are ACC's from which N-p-nitrobenzoyl derivatives were prepared, and m.p. in °C of the derivatives. cis-II, 128-129.5 (from alc.); trans-I, 185-187 (from alc.); cis-III, 141-142 (from alc.); trans-III, 164-165 (from alc.); cis-IV, 130-131 (from benzene); trans-IV, 155-156 (from benzene); cis-V, 135 (from benzene), trans-V, 137-138 (from benzene);	
CARD:		9/23 10/23	

COUNTRY : U.S.S.R.  
CATEGORY :  
ABSTRACT JOUR. : AZKhim., No. 20 1959, No. 71412  
REFFOR :  
PAGE :  
ORIG. PUB. :  
ABSTRACT : is a list of ACCs from which the N-benzoyl-O-methanesulfonyl derivatives were prepared, yleic, and derivatives m.p. in °C: cis-I, 58.7, 128-129 (from ethylacetate); trans-I, 44.2, 137-138 (from CH<sub>3</sub>COOC<sub>2</sub>H<sub>5</sub>); cis-II, 88.5, 113-114 (from benzene-petr. eth.); trans-II, 88.5, 104-105; cis-III, 88, 127-128 (from ethylacetate), trans-III, 75, 108-110 (from ethylacetate); cis-IV, 90, 120-121 (from ethylacetate-petr. eth.); trans-IV, 86, 106-107 (from ethylacetate-petr. eth.); cis-V, 11, 135-136 (from ethylacetate); trans-V, 43, 109-110 (from ethylacetate); cis-VI, 89,  
CARD: 11/23

REF ID:	134-10450-10A
CLASS:	
ADM. INFO.:	REKHA, DO. 20 1959, NO. 71412
AUTHOR:	
TYPE:	
COPY, PUB.:	
ABSTRACT:	: 119-120 (from benzene-petr. eth.); trans-VI, 95, 124-125 (from benzene-petr. eth.); cis- VII, 68, 147-148 (from ethylacetate); trans- VIII, 98, 150-151 (from ethylacetate); cis- VIII, 67, 137-139 (from ethylacetate); trans- VI; J, 68, 126-127 (from ethylacetate); cis-IX, 95, 122-123 (from ethylacetate); trans-IX, 81, 117-119 (from ethylacetate); cis-X, 42, 114-115 (from $\text{CH}_3\text{COOC}_2\text{H}_5$ -petr. eth.); trans-X, 71, 100-101 (from $\text{CH}_3\text{COC}_2\text{H}_5$ ). FO's were ob- tained by boiling 0.002 moles of N-benzoyl-ACO methanesulfonates with 0.25 g of dry $\text{CH}_3\text{COOK}$ in 150 ml of alcohol for 16 hours.
CARD:	12/23

COUNTRY	: Ussr (Soviet Union)
CATEGORY	:
ABS. JOUR.	: AZKhim., No. 20 1959, No. 71412
EDITOR	:
TRANSL.	:
ORIG. PUB.	:
ABSTRACT	: Listed below are the oxazolines prepared by this above reaction, their . . in °C/mm and mp for their picrates in °C: cis-2-phenyl-4, 5-pentamethylene- $\Delta^2$ -oxazoline (cis-XI), 115- 117/0.35, 145-146 (from alcohol); trans-XI, 135/0.9, M.P. 54-55° (from petr. eth.) and 169-171 (from alc.). cis-2-phenyl-4,5-hexa- methylene- $\Delta^2$ -oxazoline (cis-VII), 121-125/0.3, M.P. from petr. eth. 50-51.5°, 163-165 (from alc.); trans-XII, 125/0.3, 123-169 (from alc.); cis-2-phenyl-4,5-hexamethylene- $\Delta^2$ -oxazoline (cis-XIII) 132-134/0.15, 155-156 (from alc.); trans-XIII, 110-114/0.06, 155-156 (from alc.);
CARD:	13/2

1. COUNTRY : Czechoslovakia  
2. CITY : :

3. JOURNAL : RERhim., No. 20 1959, No. 71412

4. AUTHOR :  
Title :  
Address :

5. CRIM. PUB. :

ABSTRACT : cis-2-phenyl-octamethylene- $\gamma^2$ -oxazoline  
(cis-XIV), 138/0.4, 154-155 (from alc.);  
trans-XIV, (77% yield), 132-138/0.1, 150-151  
(from alc.); cis-2-phenyl-4,5-nonamethylene- $\gamma^2$ -oxazoline (cis-XV), 130-135/0.05, m.p.  
39-40° (from petr. eth.), 174-175 (from alc.);  
trans-XV, -, m.p. from petr. eth. 56-57,  
143-144 (from alc.); cis-2-phenyl-4,5-decamethylene- $\gamma^2$ -oxazoline (cis-XVI); -,  
m.p. from petr. eth. 54-55, 163-164 (from  
alc.); trans-XVI, -, m.p. from petr. eth.  
115-116, 167-168 (from alc.); cis-2-phenyl-4,  
5 undecamethylene- $\gamma^2$ -oxazoline (cis-VII), -,

CARD: 14/23

COUNTRY : Czechoslovakia  
CITY :

ACT. NO.: RZKHM., No. 20 1959, Ic. 71-12

AUTHOR :  
INST. :  
TITLE :

ORIG. PUB. :

ABSTRACT : N-benzoyl-2-ethoxyoctylamine, m.p. 153-165° (from benzene) is formed at the same time as trans-N-benzoyl-V and C<sub>10</sub>H<sub>21</sub>ON(X) are formed together with trans-C<sub>14</sub>H<sub>23</sub>ON(X). Substance XX consists apparently of a mixture of N-benzoyl-cyclodecylamines and forms a cyclodecanone [semicarbazone; m.p. 201-202°], upon boiling with 15% HCl. FO's are characterized by the melting points of known specimens obtained via condensation of ACO with benziminoethers. At the reaction conditions of method D (-15°C) trans-N-benzoyl-

DATE: 16/23

1. COUNTRY : Czechoslovakia

2. CITY : Prague

3. ADDRESS : PRAGUE, No. 20 1959, No. 71417

4. SOURCE :

5. DATE :

6. PAGE, PPR. :

7. INFORMATION : cyclododecane (22% yield, m.p. fr. benzene 170°),  $\text{CIX}^{\text{II}}$ , from cis-N-benzoyl-V. Upon boiling with an alcoholic solution of  $\text{CH}_3\text{COOK}$  VII is transformed into cis-XIV. Cis-XIV (72% yield) is formed "via trans-N-benzoyl-V. cis-XVI (37% yield) may be obtained from cis-N-benzoyl-VI. Cis-VI may be also made from trans-N-benzoyl-VII (65% yield). Trans-VII (74.5% yield) is prepared from cis-N-benzoyl-VII. Cis-VII (97.5% yield) is obtained from trans-N-benzoyl-VIII (-20°C). Cis-VIII (92.7% yield) is obtained from trans-N-benzoyl-IX. Trans-XIX (81.5% yield)

8. REF:

18/23

b

COUNTRY : Czechoslovakia  
CATEGORY :

ABD. JOUR. : RENAISS., NO. 20 (1950), NO. 71412

ANSWER?

ORIG. PUB. :

**ABSTRACT :** is obtained from cis-N-benzoyl-X. Cis-XIX (73% yield) is made from trans-N-benzoyl-X. 0.11 g of trans-N-benzoyl-V, in dioxane, saturated with HCl (3 days, ~20°) yielded 0.09 g of trans-C-benzoyl-2-aminocyclodecanol chloride (Cl), (XXXIII) m.p. 197-198° (from chloroform-ether). Cis-N-benzoyl-V remains unchanged at the same conditions. 12.9 grams of a mixture of cis- and trans-N-benzoyl-V (1:1) yields, by analogous means, 4.85 grams of XXXIII, from which by reverse migration of  $C_6H_5CO$  group were formed 3.2 grams of trans-N-benzoyl-V. By analogy with XIII

CARD: 1302

5-1

trans-*t*-butenyl-*DL*-Si, almost quantitative,

20/23

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-09E13704

G-2

COUNTRY : Czechoslovakia  
CATEGORY :

ABS. JOUR. : AZKhia., no. 20 1960, 56. 71<sup>b</sup>12

ABSTRACT :

ORIG. PUB. :

ABSTRACT : 162-183; trans-0-p-nitrobenzoyl-IX-Cl,  
almost quantitative, 163-185; trans-0-p-  
nitrobenzoyl-X-Cl, 100, 189-191. From 3.7  
grams of cyclodecanol-2-one-oxime (XXIV-  
ketone), (m.p. from ethylacetate and benzene  
102-103°) were obtained upon the reduction  
with Na in n-C<sub>4</sub>H<sub>9</sub>OH, together with a small  
amount of a mixture of cis- and trans-V, 1.89 grams  
of cyclodecylamine [b.p. 102-110°/8-10 mm,  
chlorides m.p. 174-175° (from alc.-ether);  
picrates m.p. 234-236° (from alc.)], N-benzoyl  
derivatives, m.p. 142-144° (from iso-C<sub>3</sub>H<sub>7</sub>OH).

CARD: 21/23

16

CARD: 21/23

SICHER, J. Czechoslovakia

COUNTRY :

AB5. JOUR. : RZhKhim., No. 20 1959, No. 71413

AUTHOR : Svetecova, Miroslav; Sicher, Jiri  
FIRST. : Not given

TITLE : Stereochemical Studies II. Hoffmann Dissoziation of Quaternary 2-aminoacyclonol Bases of Large and Medium-Sized Rings. Determination of Aminoacyclonol\*  
ORIG. PUB. : Cech. Listy, 1953, 22, #\*, 1956-1955. Collect  
exptl. chem. commun. 1953, 23, #\*, 1950-1950

ABSTRACT : Cis- and trans-2-aminoacylciccanol (cis- and trans-1), cis- and trans-2-aminoacyclo-  
dodecanol (cis- and trans-II), cis- and trans-2-aminoacyclotridecanol (cis- and trans-III) and cis- and trans-2-aminoacyclohexadecan-  
ol (cis- and trans-IV) were transformed into the corresponding amino bases, and  
were then subjected to Hoffmann's degradation at 20-170°/12-14 mm. Trans-I was thus  
transformed into cis-1,2-epoxycyclododecane  
(cis-V), 1.71 gram yield from 4.14 g of  
ammonium salt, b.p. 94-96°/10 mm]. Trans-II  
yielded cis-1,2-epoxycyclododecane (cis-VI)

CARD: 1/7

13

\* Structure.

G-2

COUNTRY : Czechoslovakia

"APPROVED FOR RELEASE: 03/14/2001 CIA-RDP86-00513R001550420009-5"

AB5. JOUR. : RZhKhim., No. 20 1959, No. 71413

AUTHOR :  
FIRST. :  
TITLE :

ORIG. PUB. :

ABSTRACT : [91% yield, b.p. 130-133°/15mm]. Trans-III  
gave cis-1,2-epoxycyclotridecanol (cis-VII),  
[84% yield, b.p. 141-142°/15 mm]. Trans-IV  
led to the formation of cis-1,2-epoxycyclo-  
hexadecane (cis-VIII) (84% yield, b.p.  
121-122°/0.4 mm). Cis-V was transformed  
(over acid Al<sub>2</sub>O<sub>3</sub>) into trans-1,2-cyclododecanediol  
(trans-IX), m.p. 45-50° (from petr. eth.)  
while cis-VI, cis-VII and cis-VIII, upon 1  
hour boiling with 80% dioxane in the pres-  
ence of a few drops of HClO<sub>4</sub> were also trans-  
formed into trans-1,2-cyclododecanediol,

CARD: 2/7

		3-2
COUNTRY	:	Czechoslovakia
CATEGORY	:	
ABC. JOUR.	:	RZdru., No. 20 1950, No. 71413
AUTHOR	:	
INST.	:	
TITLE	:	
ORIG. PUB.	:	
ABSTRACT	:	benzene-petr. eth.) and cis-XII, m.p. 113-114° (from petr. eth.). The results obtained prove that the initial AC has a cis-configuration. In addition they show that cis-1,2-substitutions of 10- and 12-membered rings transition into antiparallel configuration, is difficult. The latter is necessary for the formation of trans-three-membered rings. Trans-oxides are formed easily only from the ammonium bases of large ring AC. The structure of cis- and trans-XI (not described before) was elucidated on the basis of variations of the rates of

CARD: 5/7

COUNTRY	: CZECHOSLOVAKIA
CATEGORY	: Organic Chemistry. General and Theoretical
	problems of Organic Chemistry
AHC. JCUR.	: Problems of Organic Chemistry, No. 23, 1959, No. 82175
AUTHOR	: Svetloda, M.; Jonas, J.; Sicher, J.
INSTR.	: Stereochemical Studies. X. Effects of Configuration and Size of Ring upon Dissociation
TITLE	Constants of 2-aminocyclanols
ORIG. PUB.	: Chem. Listy, 1958, 52, No 8, 1596-1602; Collect. Czechoslov. Chem. Commun., 1958, 23,*
ABSTRACT	: The values of dissociation constants of pK'a cis- and trans-2-aminocyclanols, derivatives of cyclopentane down to cyclotridecane, as well as of derivatives of cyclohexadecane and cycloicosane were measured. Dependence of the dissociation constant of 2-aminocyclanols on the ring dimension is similar to the corresponding dependence discovered earlier (Pre-

\*No 8, 1551-1558

CARD:

1/8

COUNTRY :  
CATEGORY :

ABS. JOUR. : RZKhim., No.23 1959, No. 82175

AUTHOR :  
INST. :  
TITLE :

ORIG. PUE. :

ABSTRACT cont'd. : presence of intramolecular hydrogen bonds, the formation of which is influenced primarily by reciprocal remoteness of HO- and NH<sub>2</sub>- groups. The value of  $\Delta pK'a$  [ $pK'a(cis) - pK'a(trans)$ ] is a relative measure of this remoteness. The values of  $\Delta pK'a$  show that in rings which are smaller than nine-membered ones, cis-isomer forms a hydrogen bridge easier than trans-isomer. Beginning with cyclononane the hydrogen bridge is formed easier in trans-

CARD: 3/8

4-5

COUNTRY :  
CATEGORY :

ABB. JOUR. : RZhKhim., no. 23 1959, no. 32175

AUTHOR :  
EDT. :  
TITLE :

CRFG. PUB. :

ABSTRACT : isomer. Maximal remoteness of HO- and NH<sub>2</sub>- groups is attained in 12-membered ring. The hydrochlorides (HCl) of 2-aminocyclanols of

the general formula  $\text{CH}_3\text{CH}(\text{CH}_2)_p\text{CH}_2\text{OH}\text{NH}_2 \cdot \text{HCl}$  (I) were prepared from alcohol solutions of 2-aminocyclanols by the addition of ether solution of HCl and by recrystallization from the mixture of  $\text{C}_2\text{H}_5\text{OH}$  + ether or iso- $\text{C}_4\text{H}_9\text{OH}$  + ether (p in I, m.p. in °C cis- and trans-I,

CARD: L/8

COUNTRY :  
DEPLMGR :  
ACQ. JOUR. : RSKhIm., No.23 1959, No. 82175

AUTHOR :  
INST. :  
TITLE :

CHG. PUB. :

ABSTRACT : <sup>cont'd.</sup> 2K'a cis- and trans-1, are given): 5, --, --,  
9.11, 8.85; 6, --, --, 9.22, 9.07; 7, 177-179,  
115-117, 9.33, 9.30; 8, 166-167, 126-129,  
9.41, 9.37; 9, 170-171, 160-161, 9.36, 9.38;  
10, 126-130, 176-177, 9.17, 9.23; 11, 130-132,  
197-198, 9.11, 9.16; 12, 201-202, 226-230,  
8.99, 9.10; 13, 191-192, 213-215 (from also-  
hol), 8.98, 9.07; 14, --, --, \*8.94, 8.99; 16,  
217-218, 134-185, 8.88, 8.90; 20, 173-174,  
139-141, 8.35, 8.86. The following were also  
\*8.93, 8.98; 15, --, --,

CA 0:

5/8

G-3

COUNTRY :  
CATEGORY :

ABS. JOUR. : RZKhim., No. 23 1959, No. 82175

AUTHOR :  
PLAT. :  
TITLE :

ORIG. PUB. :

ABSTRACT : determined: pK'a for cis- and trans-forms of  
cont'd. N-methyl I ( $p=6$ , 9.47, 9.15) and N,N-dime-  
thyl-I ( $p=6$ , 9.14, 9.03;  $p=10$ , 8.9, 8.89).  
The values of pK'a were determined by the  
titration of  $4 \cdot 10^{-3}$  M solutions of  $*(\text{CH}_3)_4\text{NOH}$   
at 20° in an  $\text{N}_2$  atmosphere. Through the methy-  
lation of 1.15 g of trans-2-aminocyclohexane  
(14 hrs of boiling with 5 ml of 90%  $\text{HgCO}_3\text{H}$  and  
5 ml of 40%  $\text{CH}_2\text{O}$ ), trans-2-dimethylaminocyclo-  
hexane (II) is obtained;  $\text{HC}$ , m.p. 183-

DATE:

6/8

\*I in 80% aqueous methyl ether of  
methyleneglycol with 0.1 M solu-  
tion of

COUNTRY :  
CATEGORY :  
AEG. JOUR. : RZKhim., No.23 1959, No. 62175  
AUTHOR :  
INST. :  
TITLE :  
ORIG. PUB. :  
ABSTRACT cont'd. : 184.5° (from alcohol); picrate, m.p. 146-  
148.5°. Analogously to II, HC of cis-2-dime-  
thylaminocyclohexanol, m.p. 182-183°, is ob-  
tained from the cis-compound. Similarly to II  
(40 hrs of boiling), from trans-2-aminocyclo-  
decanol, trans-2-dimethylaminocyclohexanol,  
yielding 92%, b.p. 97°/1 mm, is obtained; HC,  
m.p. 127-129° (from alcohol-ether); picrate,  
m.p. 109-110 (from alcohol). Similarly, from  
the cis-compound, cis-2-dimethylaminocyclo-

CARD: 7/8

JIRI SICHER

5

Změny 44cLj

*Stereochemical studies. XI. Synthesis of *cis*- and *trans*-2-aminocyclotetradecanol<sup>1</sup>) and 2-aminocyclopentadecanol. Miroslav Svoboda, Milos Tichý, and Jiri Sicher (Českoslov. akad. věd, Prague). Chem. listy 52, 1051-8 (1958); cf. C.A. 53, 1185g.*—Hydrogenating a soln. of 100 g. cyclotetradecanol-2-one oxime in 1 l. EtOH over Adams catalyst gives 21.85 g. *cis*-2-aminocyclotetradecanol (I), m. 120-1° (C<sub>4</sub>H<sub>6</sub>, AcOEt, EtOH); HCl salt, m. 223-4° (EtOH-Et<sub>2</sub>O). Boiling 18 hrs. a soln. of 18.0 g. *trans*-2-phenyl-4,5-dodecamethylene-Δ<sup>4</sup>-oxazoline in 250 ml. EtOH and 200 ml. concd. HCl gives 13.2 g. *trans*-2-aminocyclotetradecanol (II), m. 108-8° (petr. ether); HCl salt, m. 215-16° (EtOH-Et<sub>2</sub>O). Similarly is obtained *cis*-2-aminocyclopentadecanol, m. 89-90° (petr. ether, ligroine); HCl salt, m. 220-2.5° (EtOH-Et<sub>2</sub>O). *trans*-2-Aminocyclopentadecanol, obtained in 82% yield from *trans*-2-phenyl-4,5-tridecamethylene-Δ<sup>4</sup>-oxazoline, m. 93-3.5° (petr. ether); HCl salt, m. 192-3° (EtOH-Et<sub>2</sub>O). Adding 16.0 g. NaOH in 250 ml. H<sub>2</sub>O to a soln. of 18.2 g. I in 500 ml. C<sub>4</sub>H<sub>6</sub> and then 14.8 g. BrCl portionwise with stirring and cooling gives 25.7 g. *cis*-2-benzamidocyclotetradecanol (III), m. 183-4° (EtOH). Similarly are obtained: 88.5% *trans*-2-benzamidocyclotetradecanol, m. 170-1° (EtOH); 90% *cis*-2-benzamidocyclopentadecanol, m. 142-2.5° (Me<sub>2</sub>CO); 83% *trans*-2-benzamidocyclopentadecanol, m. 143-3.5° (Me<sub>2</sub>CO-EtOH). Acylation according to the method described in C.A. 53, 1185g, yields *cis*-2-(*p*-nitrobenzamido)cyclotetradecanol, m. 194.5-5.0° (EtOH), *trans*-2-(*p*-nitrobenzamido)cyclotetradecanol, m. 192-3° (EtOH), *cis*-2-(*p*-nitrobenzamido)cyclopentadecanol, m. 162.5-3.5° (EtOH).

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and *trans*-2-(*p*-nitrobenzamido)cyclopentadecanol, m. 181-2° (EtOH). A cooled soln. of 23.2 g. III in 400 ml. C<sub>4</sub>H<sub>6</sub>N kept with 16 g. MeSO<sub>3</sub>Cl at 0° overnight, the mixt. dild. with ice H<sub>2</sub>O, the crystals filtered off, washed, and dried give 28.0 g. *cis*-2-benzamidocyclotetradecyl methanesulfonate (IV), m. 148-9° (AcOEt). Similarly are obtained: 87% *trans*-2-benzamidocyclotetradecyl methanesulfonate, m. 129-30°; 100% *cis*-2-benzamidocyclopentadecyl methanesulfonate, m. 120-1°, and 91.5% *trans*-2-benzamidocyclopentadecyl methanesulfonate, m. 107-8°. Heating 30 hrs. a soln. of 26.0 g. crude IV and 20 g. anhyd. AcOK in 1800 ml. EtOH to 96° in a glass autoclave, distg. the EtOH, and extg. the product with Et<sub>2</sub>O gives 19.0 g. *trans*-2-phenyl-4,5-dodecamethylene-Δ<sup>4</sup>-oxazoline, b.p., 100-200°, m. 62-3° (petr. ether); picrate, m. 158-60° (EtOH). Analogously are obtained: 72% *cis*-2-phenyl-4,5-dodecamethylene-Δ<sup>4</sup>-oxazoline, m. 95-6° (petr. ether) [picrate, m. 160-1° (EtOH)]; 95.8% *trans*-2-phenyl-4,5-tridecamethylene-Δ<sup>4</sup>-oxazoline, b.p., 180-90°, m. 47-8° (petr. ether) [picrate, m. 149-50° (MeOH)], and 91.6% *cis*-2-phenyl-4,5-tridecamethylene-Δ<sup>4</sup>-oxazoline, b.p., 195-202°, m. 68.5-9° (petr. ether); picrate, m. 138.5-7° (MeOH). When the N-acyl derivs. of the above aminocyclanols are subjected to acyl-migration (N → O) by dissolving 150 mg. of the compd. in 4 ml. dioxane satd. in the cold with HCl, the mixt. kept at room temp. overnight, dild. with 5 ml. dry Et<sub>2</sub>O and allowed to stand overnight, the soln. contg. the *cis* isomers remains unchanged, whereas the *trans* isomers yield cryst. HCl salts of the following compds.: *trans*-2-benzoyloxycyclotetradecylamine, m. 171-3°, *trans*-2-(*p*-nitrobenzoyloxy)cyclotetradecylamine, m. 173-5°, *trans*-2-benzoyloxy-cyclopentadecylamine, m. 106.5-7°, and *trans*-2-(*p*-nitrobenzoyloxy)cyclopentadecylamine, m. 170-81°.

L. J. Urbánek

*SICHER, J.*

Stereochemical studies. XVII. Dissociation constants of cyclohexanecarboxylic acids<sup>7</sup> and cyclohexylamines and conformational equilibria. M. Tichý, J. Jonáš, and J. Sicher (Čsl. akad. věd, Prague). Collection Czechoslov. Chem. Commun. 24, 3434-41 (1959); cf. C.A. 53, 21161i. The first energy difference between the axial and equatorial form was found to be  $1.6 \pm 0.3$  kcal. for the free, and  $2.2 \pm 0.3$  kcal. for the ionized carboxyl group in the following acids: cyclohexanecarboxylic (I), *cis*- and *trans*-4-methylcyclohexanecarboxylic (II, III), *cis*- and *trans*-3-*tert*-butylcyclohexanecarboxylic (IV, V), and *cis*- and *trans*-4-*tert*-butylcyclohexanecarboxylic (VI, VII). The calcs. were made on the basis of measurements of the dissocn. consts. in 80% Methyl Cellosolve using IV, V, VI, and VII as "axial and equatorial standards." Similar, though not so conclusive, data were collected from cyclohexylamines: cyclohexyl (VIII), *cis*- and *trans*-4-methylcyclohexyl (IX, X), and *cis*- and *trans*-4-*tert*-butylcyclohexyl (XI, XII). I, from  $\text{CO}_2$  and  $\text{C}_6\text{H}_5\text{MgBr}$ , m.  $31-2^\circ$  (petr. ether at  $-50^\circ$ ). Hydrogenation of  $p\text{-MeC}_6\text{H}_4\text{CO}_2\text{H}$  in AcOH (over PtO<sub>2</sub>) at  $45^\circ$  and atm. pressure afforded II, b<sub>d</sub>  $130.0^\circ$ , m.  $28-9.5^\circ$ .  $p\text{-MeCC}_6\text{H}_4\text{Ac}$  (78 g.) added with stirring at  $60^\circ$  to a NaOBr soln. prep'd. from 144 g. NaOH and 270 g. Br in 1200 ml. H<sub>2</sub>O, the temp. allowed to rise to  $95^\circ$ , kept at  $95^\circ$  for 30 min., the mixt. dild. with H<sub>2</sub>O, extd. with Et<sub>2</sub>O, treated with SO<sub>2</sub>, acidified with HCl, and the product filtered and crystd. gave 62 g. (80%)  $p\text{-MeCC}_6\text{H}_4\text{CO}_2\text{H}$ , m.  $165-6^\circ$  (ligroine). This (10 g.) hydrogenated over PtO<sub>2</sub> in AcOH

gave 3.2 g. VI, m.  $117.5-18^\circ$  (from EtOH, C<sub>6</sub>H<sub>6</sub>, and AcOEt, successively); Me ester (from VI and CH<sub>3</sub>N<sub>3</sub>) b<sub>d</sub>  $102-3^\circ$ , n<sub>D</sub><sup>20</sup> 1.4557. Combined mother liquors from the crystn. of VI evapd. to dryness, dissolved in 80 ml. (CH<sub>3</sub>OH)<sub>2</sub>, refluxed 7 hrs. with 20 g. KOH, the mixt. dild. with 250 ml. H<sub>2</sub>O, acidified with HCl, and the sepd. product crystd. from C<sub>6</sub>H<sub>6</sub> gave 3.4 g. VII, m.  $175.5-6^\circ$ . Hydrogenation of 40 g.  $p\text{-AcNH}_2\text{C}_6\text{H}_4\text{Me}$  in 100 ml. AcOH over 4 g. PtO<sub>2</sub> at  $65^\circ$ , distn. of the product at  $105-9^\circ$  at 0.3 mm., and crystn. (3 from cyclohexane, 2 from C<sub>6</sub>H<sub>6</sub>) gave 2.3 g. *N*-acetyl deriv. of X, m.  $140.5-1^\circ$ ; refluxing with HCl afforded 65% X, b.  $145-6^\circ$ , n<sub>D</sub><sup>20</sup> 1.4509; HCl salt m.  $260.5-1.5^\circ$  (EtOH-Me<sub>2</sub>CO); Bz deriv. m.  $180-0.5^\circ$  (aq. EtOH). From the mother liquors, after the crystn. of X, 2.5 g. *N*-acetyl deriv. of IX, m.  $98.5-9.5^\circ$  (cyclohexane and C<sub>6</sub>H<sub>6</sub>) was obtained. Hydrolysis gave IX, b.  $146-7^\circ$ , n<sub>D</sub><sup>20</sup> 1.4583; HCl salt m.  $233-4^\circ$  (EtOH-Me<sub>2</sub>CO); Bz deriv. m.  $130-0.5^\circ$  (EtOH). VI (2.26 g.) in 30 ml. concd. H<sub>2</sub>SO<sub>4</sub> and 15 ml. CHCl<sub>3</sub>, treated at  $40-5^\circ$  during 30 min. with 2.7 g. Na<sub>2</sub>N<sub>3</sub> with stirring, heated for 30 min. at  $50^\circ$ , the mixt. poured on ice, the aq. layer extd. with Et<sub>2</sub>O, alkalinized with KOH, the liberated base extd. with Et<sub>2</sub>O, and the ext. dried (MgSO<sub>4</sub>) and distd. gave 1.4 g. (73.7%) XI, b<sub>d</sub>  $70-8^\circ$ , n<sub>D</sub><sup>20</sup> 1.4670; HCl salt m.  $280-1^\circ$  (H<sub>2</sub>O); Bz deriv. m.  $157-7.5^\circ$  (EtOH).

Similarly VII was converted to 60.3% XII, b<sub>d</sub>  $77-8^\circ$ , n<sub>D</sub><sup>20</sup> 1.4648; HCl salt m.  $311-12^\circ$  (EtOH); Bz deriv. m.  $180-0.5^\circ$  (Me<sub>2</sub>CO). The dissocn. consts. in 80% Me cellosolve at  $20^\circ$  (pK<sub>a</sub>) for I through XII were: 7.43, 7.06, 7.42, 7.44, 7.06, 7.91, 7.43, 9.51, 9.44, 9.50, 9.24, 9.50.

M. Hudlický

SICHER, J.; RAJSNER, M.; RUDINGER, J.; ECKSTEIN, M.; SORM, F.

Amino acids and peptides. XXVIII. Synthesis of threo- and erythro-dl- $\alpha,\gamma$ -diamino- $\beta$ -hydroxybutyric acid ( $\gamma$ -aminothreonine and  $\gamma$ -amino-allothreonine). In English. Coll.Cz.Chem. 24 no.11:3719-3729 N '59.  
(EPAI 9:5)

1. Department of Organic Synthesis, Institut of Chemistry, Czechoslovak Academy of Science, Prague. 2. On leave of absence from the Medical Academy, Krakow, Poland (for Eckstein).

(Amino acids) (Peptides) (Allothreonine) (Amino group)  
(Threonine)

SICHER, J.; SIPOS, F.; JONAS, J.

Stereochemical studies. XVIII. Synthesis and dissociation constants  
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no.1:262-275 Ja '61. (EEAI 10:9)

1. Institute of Organic Chemistry and Biochemistry, Czechoslovak  
Academy of Science, Prague.

(Stereochemistry) (Cycloheptanedicarboxylic acid)  
(Cyclooctanedicarboxylic acid.)

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4-t-butylcyclohexanols; synthesis of the four stereoisomers. Coll  
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(Aminobutylcyclohexanol)

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The chromic acid oxidation of cycloparaffins; correlation between reactivity and thermochemical strain, and notes on reaction mechanism. Coll Cz Chem 26 no.9:2355-2369 '61.

1. Institute of Organic Chemistry and Biochemistry, Czechoslovak Academy of Sciences, Prague.

(Chromic acid) (Cycloalkanes)

3.091/02/010/021/010/009  
31-9/5101

Author: Schmitz, J., "Petry, M., Šimáček, F., Pánková, M.

Title: Conformational analysis of 2-phenylcyclohexylbenzylcarbinols; an attempt at quantitative confirmation and analysis of the part played by adjacent groups and by orientation in 1,4-difunctional derivatives of cyclohexane

PERIODICAL: Referativnyj zhurnal. Khimiya, no. 21, 1962, 116, abstract 21110 (subject: Chem. Commun., v. 26, no. 9, 1962, page 3-21). (Final summary in Russ.)

Abstract: The authors report on the thermal isomerization of the acetanilide-butenotes (I), 2-phenylcyclohexylbenzylcarbinol (II), trans-2-phenylcyclohexylbenzylcarbinol (III), N<sup>1</sup> (II), trans-2-benzamido-2-phenylcyclohexylbenzylcarbinol (IV), N<sup>1</sup> (III), cis-2-benzamido-cis-4-phenylcyclohexylbenzylcarbinol (V), N<sup>1</sup> (IV), and cis-2-benzamido-trans-4-tert-butylcyclohexylbenzylcarbinol (VI), N<sup>1</sup> (V) which takes place in absolute alcohol in benzene at 100°.

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reaction mechanism is shown in Scheme II and III. It is known<sup>1</sup> that the reaction of alkene with  $\text{AlCl}_3$  in benzene at  $-78^{\circ}\text{C}$  is reversible, which requires that the equilibrium between the reactant and intermediate, which is formed in the first step, is established rapidly. This condition is satisfied if the equilibrium is established in this step (K is equilibrium constant):

$$\text{R}-\text{CH}_2-\text{CH}_2-\text{Cl} \rightleftharpoons \text{R}-\text{CH}=\text{CH}_2 + \text{AlCl}_3 \quad K = \frac{[\text{R}-\text{CH}=\text{CH}_2][\text{AlCl}_3]}{[\text{R}-\text{CH}_2-\text{CH}_2-\text{Cl}]}$$

where  $K = 10^{10}$  at  $-78^{\circ}\text{C}$  (at  $25^{\circ}\text{C}$ ,  $K = 2 \times 10^6$  sec<sup>-1</sup> at  $60^{\circ}\text{C}$ ). It can be assumed that the equilibrium is established rapidly, since the rate of formation of intermediate is proportional to the concentration of the reactant, while III reacts with  $\text{AlCl}_3$  much faster than II (rate constant is  $2 \times 10^6$  sec<sup>-1</sup> at  $60^{\circ}\text{C}$ ). Obviously the equilibrium is established rapidly, which means that the trans- $\alpha$ -position is achieved rather quickly, probably by rotation of the molecule with other conformations. Then, polymerization of the intermediate follows, followed by closing of the ring, i.e., cyclization to form the product. The following arguments: 1) the fact that the product is formed only in one of two "cis-J", IV or V (with these conformations ethoxyvinyl cation is formed), and 2) saturated and ethoxy products are found, whereas vinyl cation does not form a carbonium ion eight form,

1. J. R. E.

...representational research...

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3156/B101

By the fact that there is a great difference in reaction rates between III and -I, and between IV and V ( $\text{M} \cdot \text{L}^{-1} \cdot \text{sec}^{-1}$ ; cis-I, 0.687; IV, 0.0017; V, 0.00017), the authors find: 1) the ratio of the reaction rates for I, II and III to the reaction rates for the corresponding bicyclic-nitrilium derivatives is 4, while the analogous substitutability of I, II and III is 1.6. This confirms that the same reaction mechanism is observed for trans-I, II and III. The authors consider the fact that etainolysis takes place faster in the case of V ( $\text{C}_6^E$ ) than in the case of IV ( $\text{C}_6^B$ ), is due to the steric stress being less reduced during the formation of the pentagonal intermediate in the case of V than in that of IV; this is because in the case of V the bulky  $\text{C}_6\text{H}_5\text{CO}_2\text{NH}-$ group becomes close to H-trigonal bipyramidal, while in the case of V it departs further from this form (by analogy with  $\text{C}_6\text{H}_5\text{CH}_2\text{CO}_2\text{NH}$  the dihedral angles between the C-H bond and the  $\text{C}_6\text{H}_5$ -bonds and  $\text{N}-\text{H}$ -bonds are taken as being  $15^\circ$  and  $105^\circ$  respectively). Analysis of the reaction rate data shows that in the case of trans-I the form of the reaction is probably "twisted" form, while the ratio between the amounts of I reacting in twist form and twist form is:

1 : 4.5/6

0/001/02/000/c21/010/069  
5136/2191

Reaction conditions: 20°C, 100 rpm.

The reaction mixture is initially homogeneous. The details of reaction are as follows: (I) + (II)  $\rightarrow$  (III) + (IV). At equilibrium state

$$K = \frac{[III][IV]}{[I][II]} = \frac{0.0292}{0.0017} = 17.2$$

It is observed that the reaction mixture is initially homogeneous. There is no change in the ratio of reactants. The quantity of II, and that of III is almost equal up to 100°C. In the infrared spectra, since the conformation of the molecule is changed, the absorption bands of the partially axial chair-conformation of the C<sub>6</sub>H<sub>5</sub>-group and CH<sub>3</sub>-group are observed. The ratio of V and VI corresponds to 90% of the diequatorial conformer. Since V, VI, and VII have almost equal energy of stabilization, no variation is noticed at 60°C. cis-II is present in the reaction mixture in 10% of the total. The positive force of the C<sub>6</sub>H<sub>5</sub>CONH<sub>2</sub> group for the stabilization of the cation of the oxazolinium ion in the reaction mixture is strong, and may be 1.5 kcal; evidently,

Reaction

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Stereochemical research...

however, as ethanolysis of the tosylate of 4-tert-butyl-2-methyl-cyclohexanol and certain preliminary considerations show, it has a higher value. To 0.01 moles of the foregoing benzamido-4-tert-butylcyclohexanol dissolved in 15 ml dry acetone, 0.02 mole of freshly distilled methane sulfonylchloride is added at -10°C; the whole is held at 0°C for 6 - 12 hr and diluted with water. The crystals are filtered off, and washed in water and petroleum ether. The percentage yields and melting points, in °C, of the substances obtained are: II 78, 93 - 94; IV 78, 126.5 - 127 (from ethyl acetate); V 46, 133 - 134 (from ethyl acetate); for the production of III, see Soviet ZIK, RZhKhim, 1962, abstract 12Zh7, melting point 140 - 141°C. When 0.01 mole of IV dissolved in 100 ml absolute alcohol is heated with 0.05 g of  $\text{KSCO}_2\text{CH}_3$  for 70 hrs at 95°C; the  $\text{KSCO}_2\text{CH}_3$  is filtered off and washed with alcohol, the filtrate evaporated in vacuo, and the residue extracted with ether and an aqueous  $\text{Na}_2\text{CO}_3$  solution; the ether extracts are washed in water and dried. Of the oil separated, 1.05 g is analyzed chromatographically on neutral  $\text{Al}_2\text{O}_3$ . 20 and 50 %, respectively, of 2-benzamido- and 6-benzamido-4-tert-butyl-cyclohexene-1 (VI and VII), 50 % of 2-benzamido-4-tert-butyl-ethoxy cyclohexane

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Stereochemical research...

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(VIII) and traces of 2-benzyl-2-methyl-cyclohexanol acetate (IX). are washed out with  $C_6H_6$  and ether. The ethanolysis of V is carried out under the same conditions, but the heating continues for 23 hrs; the oil separated amounts to 1.15 g. Fractionation on neutral  $Al_2O_3$  has shown that it consists of 25 % VII, 15 % VIII, and 5 % IX. [Abstract ends. Complete translation.

Card 6/6

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SICHER, J.; CHEREST, M.; GAULT, Y.; FELKIN, H.

no academic degrees indicated

Institute of Organic Chemistry and Biochemistry, Czechoslovak Academy  
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